

Synthesis, structural and conformational analysis of a 3×3 isomer grid based on nine Methyl-N-(pyridyl)benzamides Pavle Mocilaca, Mark Tallona, Alan J. Loughb and John F. Gallaghera

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Introduction

The focus of our research is to bridge solid state structural studies with computational (ab initio) modelling methods by exploring the influence of different functional groups and their position in semi-rigid small drug-like molecules.¹⁻³ A 3×3 isomer grid of nine Methyl-N-(pyridyl)benzamides ($C_{13}H_{12}N_2O$) as Mxx (x = para-/meta-/ortho-) has been synthesised and characterised to evaluate and correlate structural relationships from both solid-state (Table 1, Figs. 1,2,4) and ab initio calculations (Fig. 3). The effect of methyl group (Mx) and pyridine N atom (x) substitution patterns on molecular structure and conformations (syn/anti, Scheme 1B) from calculations (gas phase and solvated forms), as well as on crystal packing and conformations in solid state is explored, allowing evaluation and rationalisation of disorder and unexpected conformations observed in solid state structures



Scheme 1. A. General structure diagram of Mxx compounds; B. Possible Mxx conformations isomers

Experimental methods

Classical nucleophilic acyl substitution reactions (Schotten-Baumann) of the 4-, 3- or 2-toluoyl chlorides with 4-, 3- or 2-aminopyridines produced a series of nine Mxx compounds. The Mxo triad was synthesised under solventless conditions. Purification was accomplished by standard organic wash up and column chromatography. Single crystals were grown from chloroform or ethyl acetate. The single crystal X-ray data were collected on an Enraf-Nonius κ-CCD diffractometer at 150(1) K and for Mpo on a Bruker ApexII at the University of Toronto : θ range from 2-27.5° with 100% data coverage to 25°. Data have also been collected at 294(1) K.



Fig. 1 Molecular structure of Mpp with displacement ellipsoids drawn at the 30% probability level.



Fig. 2 The intermolecular N-H...N_{pyridine} interactions in Mpp



Table 1. Selected crystallographic data and relevant structural features for the nine Mxx isomers

| Mxx | Space group | Z' | Volume (Å ³) | R-factors | C ₆ /C ₅ N (°) | C ₆ /amide (°) | C ₅ N/amid e (°) | NN/O (Å) | Packing |
|-----|------------------|----|-----------------------------|--------------|---|------------------------------|--------------------------------|-------------|-----------|
| Мрр | $P2_l/n$ | 1 | 1038.49(12) | 0.066, 0.202 | 46.03(9) | 28.99(11) | 17.40(13) | 3.105(3) | 1D chains |
| Mmp | $P \overline{1}$ | 2 | 1056.41(16) | 0.066, 0.183 | 4.84(15) | 15.70(13) | 10.94(13) | 3.058(3) | 2D chains |
| | | | | | 5.47(12) | 19.88(10) | 17.73(9) | 3.059(3) | |
| Мор | Pc | 2 | 1123.65(12) | 0.056, 0.161 | 81.79(15) | 54.60(17) | 28.34(17) | 2.952(5) | 1D chains |
| | | | | | 81.58(14) | 54.85(15) | 27.96(16) | 2.950(5) | |
| Mpm | PI | 4 | 1071.37(8) | 0.059, 0.154 | 65.13(11) | 29.45(15) | 35.83(14) | 3.029(6) | 1D chains |
| | | | | | 64.04(10) | 29.70(15) | 34.57(13) | 3.016(5) | |
| | | | | | 1.5(2) | 24.47(16) | 24.16(16) | 3.085(6) | |
| | | | | | 1.9(2) | 24.79(16) | 24.79(16) | 3.085(5) | |
| Mmm | $P \overline{1}$ | 2 | 1079.21(9) | 0.076, 0.254 | 55.57(10) | 25.55(8) | 31.33(14) | 2.998(4) | 1D chains |
| | | | | | 65.70(10) | 36.43(10) | 29.54(14) | 3.006(4) | |
| Mom | Pca21 | 1 | 1109.94(5) | 0.041, 0.096 | 71.20(5) | 57.33(6) | 13.91(10) | 2.946(2) | 1D chains |
| Мро | $P \overline{1}$ | 1 | 525.04(5) | 0.039, 0.109 | 45.34(4) | 35.93(5) | 17.51(8) | 3.1081(15) | Dimer |
| Mmo | $P2_I/c$ | 1 | 1065.04(15) | 0.065, 0.195 | 70.55(7) | 35.99(8) | 35.05(11) | 3.106(3) | Dimer |
| Moo | $P \overline{1}$ | 1 | 546.42(6) | 0.063,0.184 | 84.79(5) | 77.28(6) | 10.93(10) | 3.076(3) | Dimer |

In silico methods

The Mxx isomer optimisation and conformational analysis giving PES diagrams was performed using ab initio calculations (B3LYP/6-311++G, corresponding B3LYP/6-311++G** studies are in progress) on isolated (gas-phase) and solvated molecules (PCM-SMD solvation model with CH₂Cl₂ or H₂O as solvents) using Gaussian03/09 together with high accuracy energy calculations (CBS-QB3) and the ΔG of solvation. Corresponding solid state structure dihedral angles were plotted in gas phase PES diagrams relative to optimised structures dihedral angles.



Fig. 3 The PES conformational analysis diagrams for the Mxx isomers optimised in the gas phase: the equivalent solid state angle is depicted as a (•), with, if applicable, assigned identification letter and partial occurrer letter and partial occurrence (%).



Fig. 4 A: The Moo hydrogen bonded dimer with symmetrical C-H... π(arene) interaction depicted for H23 only. B: The Moo 'dimer' with atoms as their van der Waals spheres

Conclusion

The primary hydrogen bonding is a_{amide} N-H...N_{pyrdine} in eight isomers but is a_{amide} N-H...O=C $_{\text{amide}}$ in **Mpm**. The N-H...N interactions form *C*(6) chains in the **Mxp** series, C(5) chains in Mmm, Mom and cyclic centrosymmetric $R^2_2(8)$ rings in the Mxo series. Of note is the short intradimer C-H... π (arene) interaction in **Moo** (Fig. 4), one of the shortest reported in a neutral organic system C...Cg = 3.3875 (18) Å, H...Cg = 2.46 Å [2.33 Å] and C-H...Cg = 167° [166°] [with normalised C-H distances]. For five of the nine isomers the torsion angle derived from *ab initio* calculations are reasonably consistent with the crystallographic data (Fig. 3). The Manti/P-syn conformation is preferred in all modelled structures. However, for Mmp, Mmm, Mom and Moo, the crystallographic torsion angles are not located in an energetically favourable conformation based on our ab initio results, adopting metastable or unstable conformations.

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