



# Organic Electrochemical Transistor Incorporating an Ionogel as a Solid State Electrolyte for Lactate Sensing



Vincenzo F. Curto<sup>a</sup>, Dion Khodagholy<sup>b</sup>, Kevin J. Fraser<sup>a</sup>, Moshe Gurfinkel<sup>b</sup>, Dermot Diamond<sup>a</sup>, George G. Malliaras<sup>b</sup>, Róisín M. Owens<sup>b</sup>, Fernando Benito-Lopez<sup>a</sup>

<sup>a</sup> CLARITY: Centre for Sensor Web Technologies, National Centre for Sensor Research, Dublin City University, Dublin 9, Ireland

<sup>b</sup> Centre Microélectronique de Provence, Ecole Nationale Supérieure des Mines de Saint Etienne, 880, route de Mimet, 13541 Gardanne, France.

contact detail: [vincenzo.curto2@mail.dcu.ie](mailto:vincenzo.curto2@mail.dcu.ie)

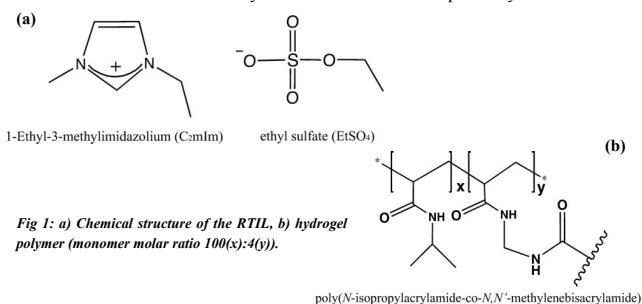
## Introduction:

Room temperature Ionic liquids (RTILs) have evolved as a new type of solvent for biocatalysis, mainly due to their unique and tunable physical properties.[1] In addition, within the family of organic semiconductor-based sensors, organic electrochemical transistors (OECTs) have attracted particular interest.[2]

Here, we present a simple and robust biosensor, based on a OECT, capable of measuring lactic acid using a gel-like polymeric materials that endow RTIL (ionogel)[3] as solid-state electrolyte both to immobilise the enzyme and to serve as a supporting electrolyte.[4] This represents the first step towards the achievement of a fast, flexible, miniaturised and cheap way of measuring lactate concentration in sweat.

## Experimental:

1-Ethyl-3-methylimidazolium ethyl-sulfate ionic liquid, [C<sub>2</sub>mIm][EtSO<sub>4</sub>], was chosen because of its miscibility with water and bio-compatibility.



The OECT is fabricated by standard lithography and it is made of 200 nm thick PEDOT:PSS film.

The hydrated RTIL mixture containing the lactate oxidase enzyme (LOx) was photo-polymerised using a UV irradiation source ( $\lambda = 365$  nm) for 1 minute.

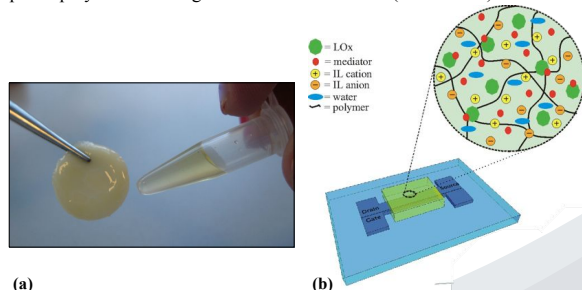


Fig 2: a) Ionogel after (left) and before polymerisation (right); b) schematic representation of the OECT device with ionogel/enzyme mixture.

## Acknowledgement

Science Foundation Ireland (SFI) under the CLARITY CSET award (Grant 07/CE/11147) and Research Career Start Programme 2010 fellowship from Dublin City University, Marie Curie Actions International Reintegration Grant (IRG) (PIRG07-GA-2010-268365) and Irish Research Council for Science, Engineering and Technology, Marie Curie International Reintegration Grant (Grant PIRG06-GA-256367 CELLTOX).

## Results & Discussion:

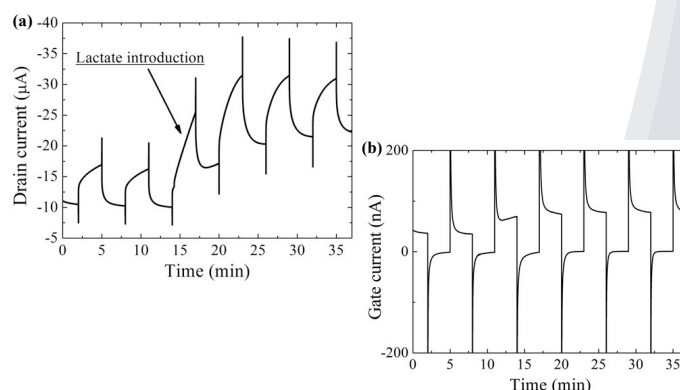


Fig 3: a) Drain current vs. time with addition of 25 mM lactate indicated by an arrow; b) corresponding gate current vs. time.

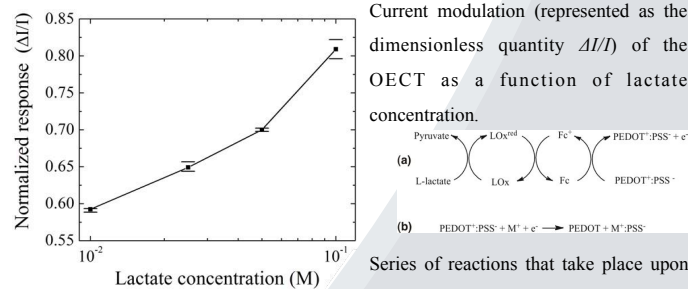


Fig 4: Normalised response of the OECT vs. lactate concentration and reactions at the gate electrode (a) and at the channel (b) of the OECT.



Fig 5: OECT with gel shown on a forearm.

## Conclusions:

We demonstrate the detection of lactate in a relevant physiological range using an OECT sensor with an ionogel solid-state electrolyte. The significance of this work for sensing applications lies in the configuration of the sensor; we show for the first time a solid state electrolyte on a flexible transistor-based biosensor.

## References

- [1] H. Zhao, J. Chem. Tech. Biotech, 2010, 85, 891-907.
- [2] Bernards, D.A., D.J. Macaya, M. Nikolou, J.A. DeFranco, Se. Takamatsu and G. G. Malliaras, J. Mat Chem, 2008, 18(1), 116-120
- [3] J. Le Bideau, L. Viau and A. Vioux, Chem. Soc. Rev., 2011, 40(2), 907-925
- [4] Khodagholy, D., V.F. Curto, K.J. Fraser, M. Gurfinkel, R. Byrne, D. Diamond, G. G. Malliaras, F. Benito-Lopez and R.M. Owens, J. Mat Chem, 2012, 22(10), 4440-4443