



Use of Light Emitting Diodes in the Visible Region to Initiate Polymerisation leading to Monolithic Stationary Phases



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Outline

- The current state of the art of monolith synthesis
- The advantages of working in the visible region
- The benefits of working with LEDs
- My research
 - □ Part I: Initiation at 660 nm
 - □ Part II: Initiation at 470 nm
- Conclusions



Initiation of monolith synthesis: State of the art

Thermal initiation
All standard moulds
Long polymerisation times
No spatial control



Photo initiation
Light transparent moulds
Short polymerisation times
Excellent spatial control



Gamma radiation initiation
All standard moulds
Short polymerisation times
No spatial control

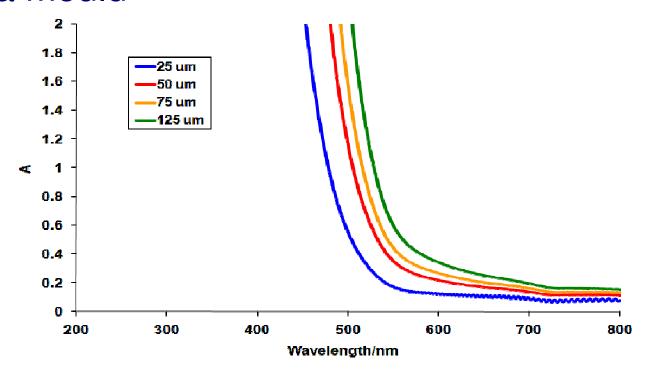
Microwave initiation
All standard moulds
Short polymerisation times
No spatial control

Electron beam initiation
All standard moulds
Short polymerisation time
No spatial control

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Advantages of working in the visible region....

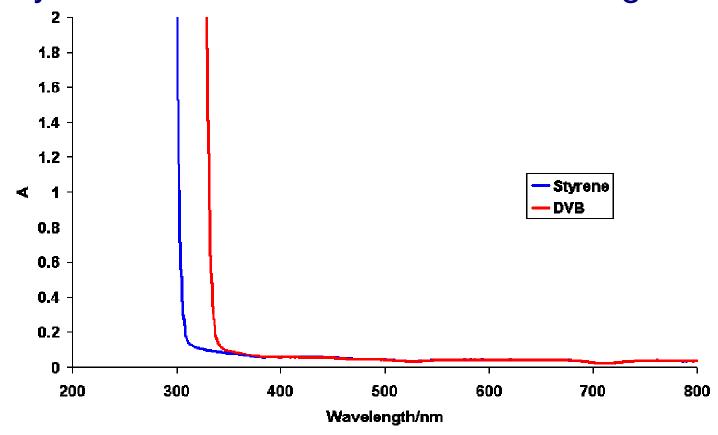
 Polyimide does not absorb strongly above 550 nm so polyimide coated capillaries can be used as a mould



b/A

Advantages of working in the visible region....

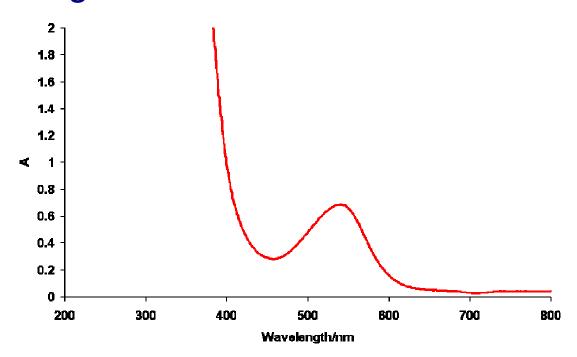
Styrenes don't absorb in the visible region



be.

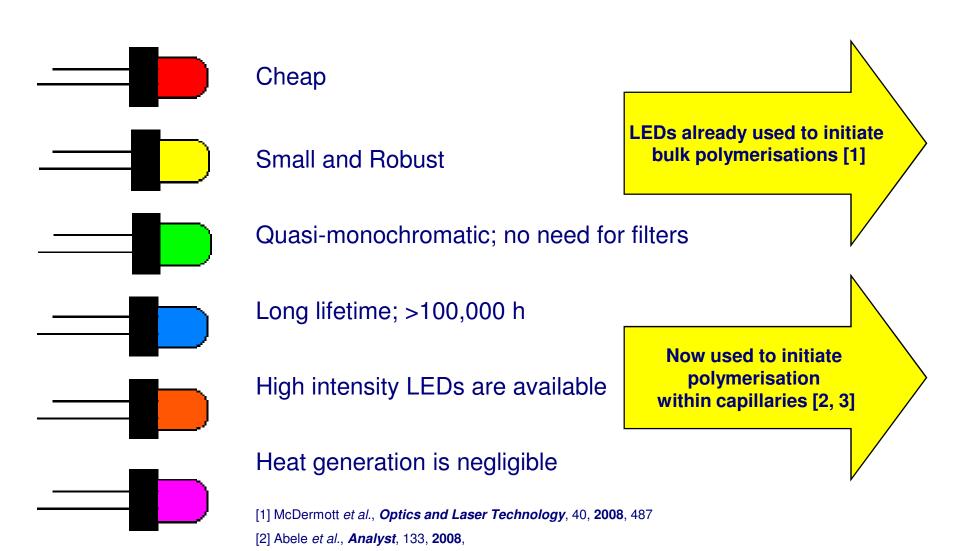
Advantages of working in the visible region....

Monomeric dyes, while absorbing strongly in the UV, usually don't absorb throughout the whole visible region





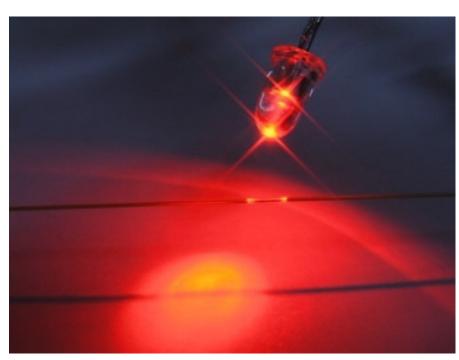
The Use of LEDs



[3] Walsh et al., Chem.Comm., 48, 2008, 6504

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Part I: Initiation at 660 nm



Walsh et al., Chem.Comm. 48, 6504-6506, 2008

 Polymerisation of methacrylate monoliths within polyimide coated capillaries with 660 nm LEDs and a cyanine/borate initiator complex

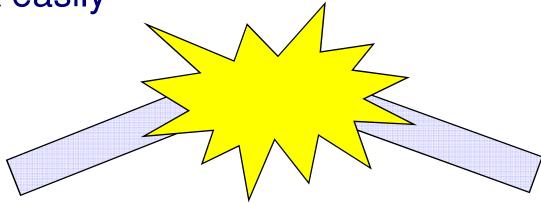


Identification of a problem

 Photoinitiated polymerisation can be done easily with UV light in PTFE coated capillary....

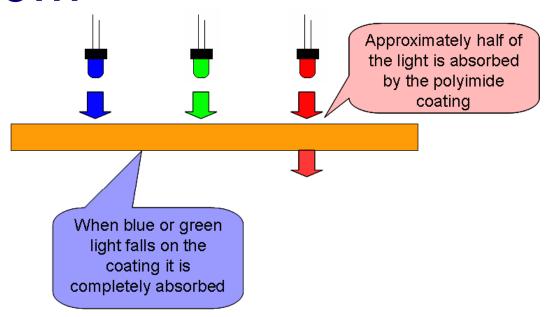


However, this capillary is quite brittle and can break easily



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Polyimide makes the capillary more durable...

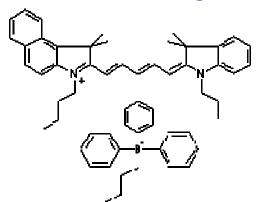


- While it becomes more durable, it also becomes less transparent
- Only in the red region of the spectrum does the light pass more easily through the capillary into the internal cavity

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Selection of the Initiator

Initiator absorbing above 550 nm is necessary



 $\lambda_{\text{max}} = 660 \text{ nm}$



Radicals generated very slowly = long polymerisation time



$$\begin{bmatrix} \mathbf{C}\mathbf{y} \end{bmatrix}^{\bullet} + \bigoplus_{\Theta} \mathbf{N}$$

$$h\nu \downarrow$$

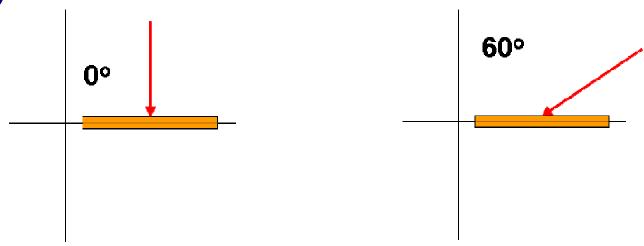
$$\begin{bmatrix} \mathbf{C}\mathbf{y} \end{bmatrix}^{\oplus} + \mathsf{MeO}^{\bullet} + \begin{bmatrix} \mathbf{N} \\ \mathbf{N} \end{bmatrix}$$

Quick reaction!

20 – 120 min



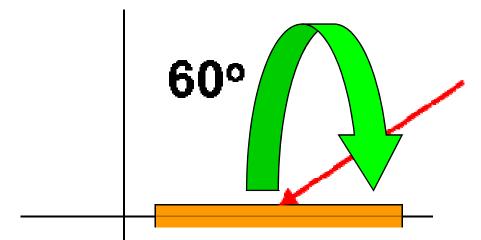
Angle of the capillary relative to the LFD



- Perpendicular position creates only one focal point
- Position at 60° creates multiple focal points
- This means more light within the cavity



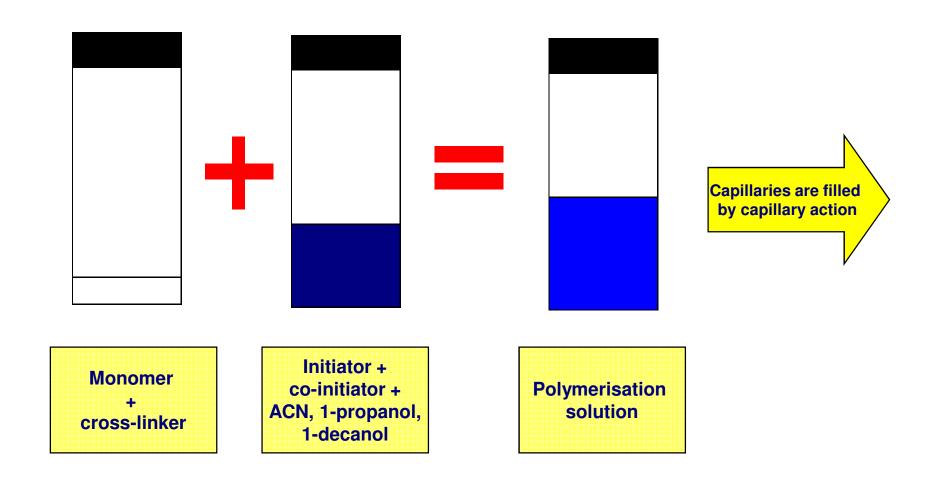
Rotation of the capillary



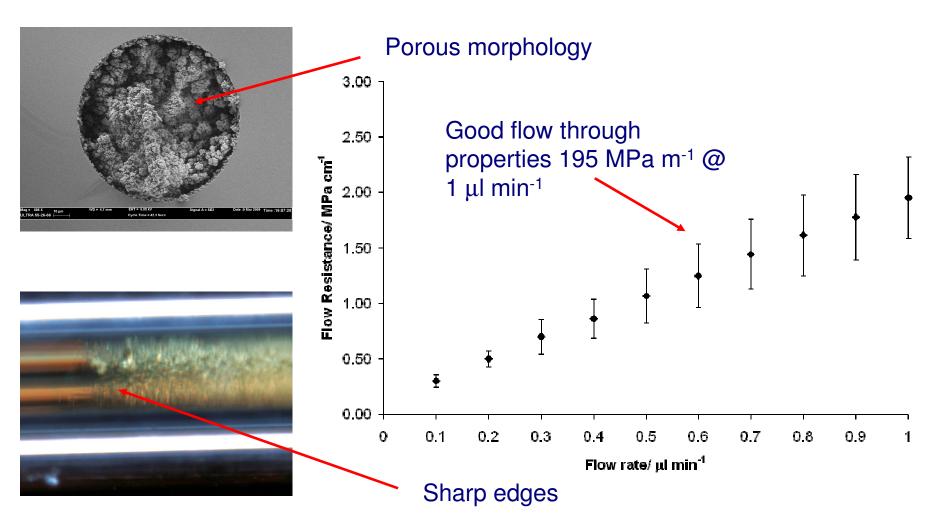
- Rotation of the capillary was found to be important
- It ensures homogeneous distribution of the monolith within the capillary



Polymerisation Conditions

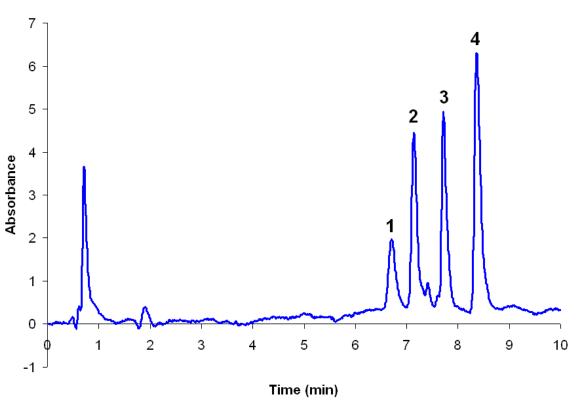


Characterisation of monoliths



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Separation of Proteins

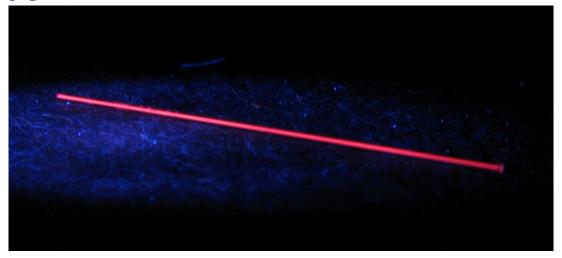


- Poly (BuMA-EDMA) monolith
- Flow rate = 1µl/min
- Detection = 210 nm
 - Gradient = 0-60% ACN in 5 min

(1) Ribonuclease A (2) Cytochrome C (3) Myoglobin, (4) Ovalbumin

Other applications of the initiator

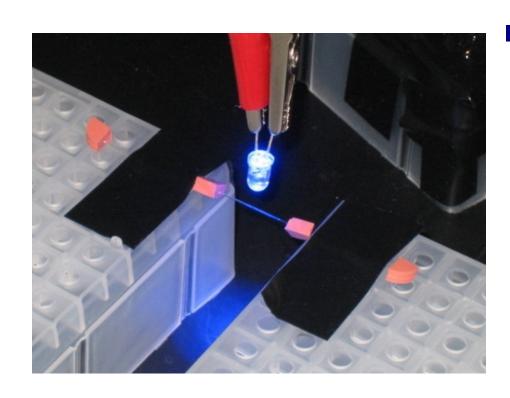
 Red light initiation system has also been used to graft chromophoric monomers onto monolithic scaffolds



Poly (butyl methacrylate-co-ethylene dimethacrylate) with a spiropyran monomer grafted into the pores. The monolith is illuminated with a 375 nm LED to show the fluorescence of the spiropyran in the pores.

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Part II: Initiation at 470 nm



Polymerisation of styrene-based monoliths within polytetrafluoroethylene coated capillaries with 470 nm LEDs and a camphorquinone/ethyl-4-dimethylamino benzoate initiator complex



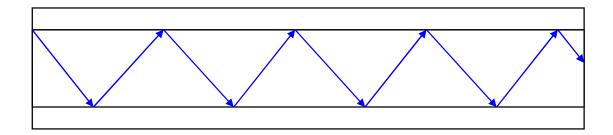
If red works, why use blue?

- Red light can be used to initiate the more reactive methacrylate monomers
- Styrenes, however, are more stable and need more energy to start to polymerise
- Styrenes also absorb in the UV region so UV is not an option
- Blue is the highest energy light in the visible region so we use 470 nm LEDs to polymerise styrene and divinylbenzene



Position of the LED

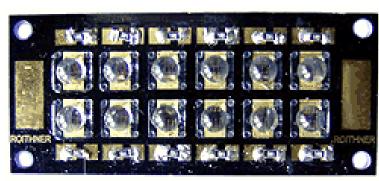
As PTFE coated capillary is a good waveguide the angle at which the LED is positioned is not important



- Effective photo-masking can be a problem as light can travel under the mask
- Rubber septa were used as photo-masks for all the blue light experiments

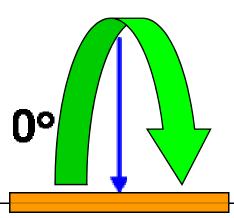


LED arrays for longer columns



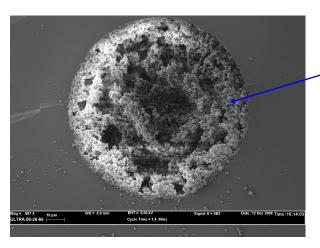
http://www.roithner-laser.com/LED HP multi chip.html

 When using the LED array, rotation is important as differences between light outputs from each LED produce column inhomogeneities



 Due to waveguiding when using single LEDs rotation is unnecessary

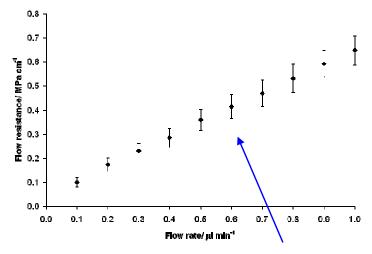
Characterisation of monoliths

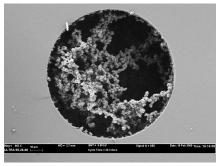


Sharp edged monolith can be obtained with LED array using photomasks

Polymerisation obtained with single low power LED

Polymerisation obtained with high-power LED array

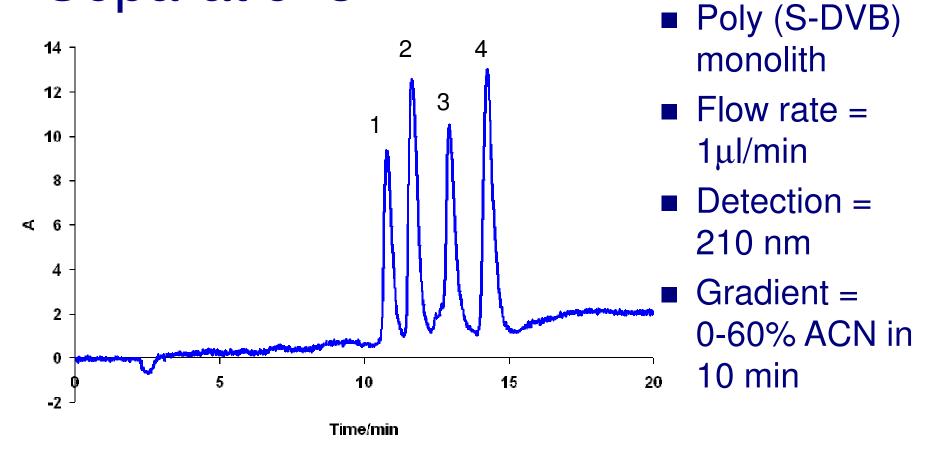




Low backpressure suggests good permeability

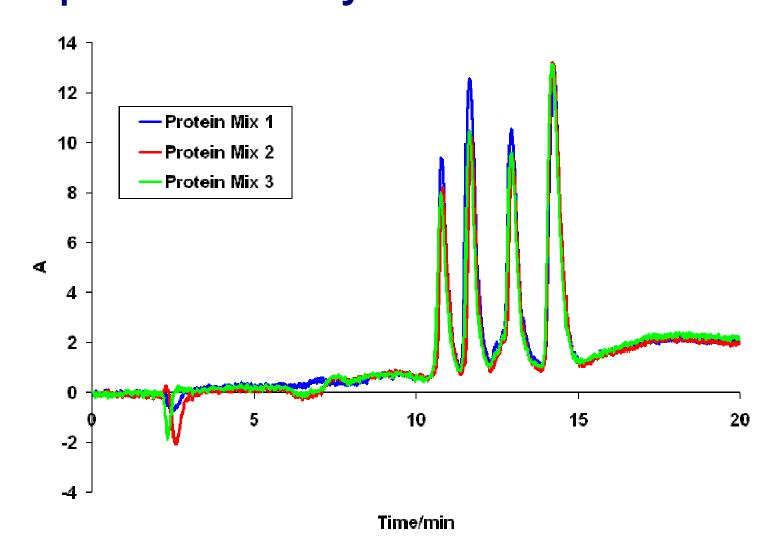


Separations



(1) Ribonuclease A (2) Cytochrome C (3) Myoglobin, (4) Ovalbumin

Reproducibility





Conclusions

- Successful synthesis of a range of methacrylate monoliths within polyimide coated fused silica capillary
- Grafting of a chromophoric spiropyran monomer onto the surface of a poly (BuMA-EDMA) monolith using red light
- Successful polymerisation of PS-DVB monoliths in PTFE coated capillary in the visible region using blue light



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