ExoMol line lists XXIV: a new hot line list for silicon monohydride, SiH

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ABSTRACT

SiH has long been observed in the spectrum of our Sun and other cool stars. Computed line lists for the main isotopologues of silicon monohydride, ^{28}SiH , ^{29}SiH , ^{30}SiH and ^{28}SiD are presented. These line lists consider rotation–vibration transitions within the ground *X* ² Π electronic state as well as transitions to the low-lying *A* ² Δ and *a* ⁴ Σ^- states. Ab initio potential energy (PECs) and dipole moment curves along with spin–orbit and electronic angular momentum couplings between them are calculated using the multireference configuration interaction level of theory with the MOLPRO package. The PEC for the ground $X^2\Pi$ state is refined to available experimental data with a typical accuracy of around 0.01 cm−¹ or better. The ²⁸SiH line list includes 11 785 rovibronic states and 1724 841 transitions with associated Einstein-*A* coefficients for angular momentum *J* up to 82.5 and covering wavenumbers up to 31 340 cm⁻¹ (λ < 0.319 µm). Spectra are simulated using the new line list and comparisons made with various experimental spectra. These line lists are applicable up to temperatures of 5000 K, making them relevant to astrophysical objects such as exoplanetary atmospheres and cool stars and opening up the possibility of detection in the interstellar medium. These line lists, called SiGHTLY, are available at the ExoMol [\(www.exomol.com\)](http://www.exomol.com) and CDS data base websites.

Key words: molecular data – opacity – astronomical data bases: miscellaneous – planets and satellites: atmospheres – stars: low-mass.

1 INTRODUCTION

Silicon hydride (SiH) is a free radical formed from the cosmically abundant elements of hydrogen and silicon, and is the simplest of the four possible silicon hydrides. Following the first experimental measurement of the $A^2\Delta - X^2\Pi$ system of SiH by Jackson [\(1930\)](#page-8-0), SiH was observed by Pearse [\(1933\)](#page-9-0) and Babcock [\(1945\)](#page-8-1) in sunspots and in the spectrum of the solar disc spectrum by Schadee [\(1964\)](#page-9-1) and Moore-Sitterly [\(1966\)](#page-8-2). These identifications were then corroborated by Sauval [\(1969\)](#page-9-2) using the coincidence method and also by Lambert & Mallia [\(1970\)](#page-8-3), who derived oscillator strengths and isotope shifts for ²⁹SiH and ³⁰SiH. Furthermore, Grevesse & Sauval [\(1970\)](#page-8-4) observed SiH in the photospheric region of the sun and also subsequently calculated oscillator strengths (Grevesse & Sauval [1971\)](#page-8-5).

SiH has been observed in late-type stars by Davis [\(1940\)](#page-8-6) and in the emission spectra of M- and S-type Mira variable stars (Merrill [1955\)](#page-8-7). SiH is also important for the modelling of M-dwarf atmospheres (Allard & Hauschildt [1995\)](#page-8-8), although of less importance than species with more pronounced spectroscopic features.

In exoplanetary and brown dwarf atmospheres, Visscher, Lodders & Fegley [\(2010\)](#page-9-3) make the prediction that SiO should be the most abundant silicon species in low-pressure environments, with silane (SiH4) the most abundant at high pressures. They also predict that SiH may be present due to equilibrium reactions of SiH with both $SiH₄$ and SiO in the presence of $H₂O$.

In the interstellar medium (ISM), given the observations of similar molecules such as SiO, CH and OH, the presence of SiH has been suggested by numerous authors (Weinreb et al. [1963;](#page-9-4) Wilson et al. [1971;](#page-9-5) Rydbeck, Elldér & Irvine [1973;](#page-9-6) Lovas [1974;](#page-8-9) Turner & Dalgarno [1977;](#page-9-7) de Almeida & Singh [1978\)](#page-8-10). In particular, Herbst et al. [\(1989\)](#page-8-11) suggested that the abundances of both silicon and hydrogen make SiH a likely candidate to be found in interstellar clouds, where it so far remains undetected.

Several theoretical studies exploring the electronic and thermodynamic properties of SiH have been made, often as parts of larger studies of silicon hydrides. The first ab initio electronic structure calculations were performed by Cade & Huo (1967) who produced a ground state potential energy curve (PEC) using the Hartree–Fock method and Slater-type orbital basis functions. Ramakrishna Rao & Lakshman [\(1971\)](#page-9-8) used the Rydberg–Klein–Rees method as well as their own modified method to calculate the PECs for the $X^2\Pi$ and $A^2\Delta$ states for SiH on the basis of the known experimental

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data. Later Meyer & Rosmus [\(1975\)](#page-8-13) used the coupled electron pair approximation method and a Gaussian-type orbital basis set to examine both the PEC and DMC of the SiH ground state. This work on the PEC for the ground state has been followed up by Shi et al. [\(2008\)](#page-9-9) and Prascher, Lucente-Schultz & Wilson [\(2009\)](#page-9-10) using the coupled-cluster method, with the latter of these studies forming part of a more general, larger study of silicon-containing hydrides. Additionally, theoretical calculations of the ground state dipole were also performed by several authors using various methods (Meyer & Rosmus [1975;](#page-8-13) Kalcher [1987;](#page-8-14) Ajitha & Pal [1999;](#page-8-15) Park & Sun [1992;](#page-9-11) Pettersson & Langhoff [1986\)](#page-9-12). The transition dipole between the *X* ${}^{2}\Pi$ and *A* ${}^{2}\Delta$ states for SiH was calculated by Larsson [\(1987\)](#page-8-16) using the CASSCF method. Buenker [\(1986\)](#page-8-17) used SiH as a test case for his theoretical study on the calculation of excited molecular states.

A considerable step forward was made by Kalemos, Mavridis & Metropoulos [\(2002\)](#page-8-18), who used the multireference configuration interaction (MRCI) method to compute 16 electronic states of SiH. More recently, Shi et al. [\(2013\)](#page-9-13) produced PECs for seven bound states of SiH and gave a new set of spectroscopic parameters using, again, the MRCI method (aug-cc-pV6Z basis set for Si and aug-ccpV5Z for H) and accounting for spin–orbit (SO) coupling using the Breit–Pauli Hamiltonian.

SO coupling in SiH has also been specifically studied by several authors (Stevens & Krauss [1982;](#page-9-14) Chang & Sun [2003;](#page-8-19) Song et al. [2008;](#page-9-15) Shi et al. [2013\)](#page-9-13); more general studies of SO coupling in diatomic molecules have been undertaken by Brown & Watson [\(1977\)](#page-8-20), Qui-Xia, Tao & Yun-Guang [\(2008\)](#page-9-16), Baeck & Lee [\(1990\)](#page-8-21) and Chang & Sun (2008) . A splitting in SiH was calculated by Wilson & Richards [\(1975\)](#page-9-17), complementing various experimental measurements (Freedman & Irwin [1976;](#page-8-23) Klynning, Lindgren & Sassenberg [1979;](#page-8-24) Cooper & Richards [1981\)](#page-8-25).

SiH is of considerable interest to the semiconductor industry and it is a by-product in the production of thin films for devices such as LCDs (Jasinski, Becerra & Walsh [1995;](#page-8-26) Turban, Catherine & Grolleau [1980,](#page-9-18) [1981;](#page-9-19) Drevillon & Toulemonde [1985\)](#page-8-27). Neutral radicals, particularly $SiH₃$ provide the most efficient growth in the chemical vapour deposition process, with SiH only occurring in comparatively negligible quantities (Robertson & Gallagher 1986). It is usually produced from SiH₄ using photolysis, radio frequency discharge, or by equilibrium reaction with fluorine (Kalemos et al. [2002\)](#page-8-18). Numerous studies of SiH within silane plasmas has been undertaken (Matsuda et al. [1980;](#page-8-28) Taniguchi et al. [1980;](#page-9-21) Kampas & Griffith [1981;](#page-8-29) Schmitt et al. [1984\)](#page-9-22) and also of thermodynamic properties such as bond strength, enthalpies and heats of formation of silicon-containing hydrides (Husain & Norris [1979;](#page-8-30) McMillen & Golden [1982;](#page-8-31) Sax & Kalcher [1991;](#page-9-23) Leroy et al. [1992;](#page-8-32) Grant & Dixon [2009;](#page-8-33) Jun et al. [2010\)](#page-8-34).

Experimentally, rovibrational transition wavelengths for SiH within the ground state were measured extensively during the 1980s (Brown & Robinson [1984;](#page-8-35) Brown, Curl & Evenson [1984;](#page-8-36) Brown, Curl & Evenson [1985;](#page-8-37) Davies et al. [1985;](#page-8-38) Betrencourt et al. [1986;](#page-8-39) Seebass et al. [1987\)](#page-9-24). However, the first recorded wavelengths for SiH were actually for the $A^2\Delta - X^2\Pi$ system starting with Jackson [\(1930\)](#page-8-0), with the most recent study by Ram, Engleman & Bernath [\(1998\)](#page-9-25) building on earlier work (Rochester [1936;](#page-9-26) Douglas [1957;](#page-8-40) Verma [1965;](#page-9-27) Singh & Vanlandingham [1978;](#page-9-28) Klynning et al. [1979\)](#page-8-24). Spectra for this system were also recorded in the gas phase as part of silane glow discharge studies (Perrin & Delafosse [1980;](#page-9-29) Washida et al. [1985;](#page-9-30) Nemoto et al. [1989;](#page-9-31) Stamou, Mataras & Rapakoulias [1997\)](#page-9-32). Additionally, the overall absorption crosssections and electronic transition moment for this system were determined by Park [\(1979\)](#page-9-33) using a shock tube.

Limited experimental work has been undertaken for other excited states; Bollmark, Klynning & Pagès ([1971\)](#page-8-41) recorded spectra around 3250 Å which they attributed to transitions involving the $B^2\Sigma^$ and $C^2\Sigma^+$ states. The $E^2\Sigma^+ - X^2\Pi$ and $D^2\Delta - X^2\Pi$ systems for SiH and SiD around 1907 and 2058 Å, respectively were recorded by Herzberg, Lagerqvist & McKenzie [\(1969\)](#page-8-42), following earlier, preliminary work by Verma [\(1965\)](#page-9-27). Finally, Johnson & Hudgens [\(1989\)](#page-8-43) located a state at 46700 \pm 10 cm⁻¹ which they classified as either ${}^{2}\Pi$ or ${}^{2}\Sigma^{+}$ using resonance-enhanced multiphoton ionization spectroscopy.

The Cologne Database for Molecular Spectroscopy (CDMS) lists 125 purely rotational transition lines within the vibrational ground state spanning the 0.2–5275 GHz range (radio and microwave spectral range) (Müller et al. 2005). Kurucz (2011) has compiled a larger line list of 78 286 transitions, last updated in 1998 for *J* up to 37.5. Oscillator strengths and Franck–Condon factors have been measured by Smith & Liszt (1971) whilst measurements of lifetimes have been recorded by Smith [\(1969\)](#page-9-36), Bauer et al. [\(1984\)](#page-8-45) and Nemoto et al. [\(1989\)](#page-9-31).

The ExoMol project aims at providing line lists of spectroscopic transitions for key molecular species which are likely to be important in the atmospheres of extrasolar planets and cool stars (Tennyson & Yurchenko [2012;](#page-9-37) Tennyson et al. $2016c$). This is essential for the continued exploration of newly discovered astrophysical objects such as exoplanets, for which there is an increasing desire to characterize their atmospheric compositions. The methodology of the line list production for diatomics is discussed by Tennyson & Yurchenko [\(2017\)](#page-9-39). ExoMol has already provided rotation–vibration line lists for the closed shell silicon-containing molecules SiO (Barton, Yurchenko & Tennyson 2013) and SiH₄ (Owens et al. [2017\)](#page-9-40). Given the astronomical interest in SiH, we present a new line list for SiH applicable for temperatures up to 5000 K.

2 METHOD

The procedure used here for the calculation of SiH line lists is the established ExoMol methodology of refining ab initio results to available experimental data. Because of the absence of a complete set of potential energy curves (PECs), dipole moment curves (DMCs), SO curves (SOC) and electronic angular momentum coupling curves (EAMC) in the literature, ab initio calculations of the four lowest-lying electronic states were performed using the program MOLPRO (Werner et al. [2010,](#page-9-41) [2012\)](#page-9-42).

The ab initio PEC (see Fig. [1\)](#page-2-0)[,](#page-2-1) SOC and EAMC curves were produced at the MRCI level of theory in conjunction with the aug-ccpwCVQZ basis sets (Dunning [1989;](#page-8-47) Woon & Dunning [1993;](#page-9-43) Peterson & Dunning [2002\)](#page-9-44) with relativistic corrections, core-correlation effects and Davidson correction taken into account. The PECs of $X^2\Pi$ and $A^2\Delta$ as well as SOCs, EAMCs between the *X*, *A* and *B* states were then refined using DUO (Yurchenko et al. [2016\)](#page-9-45) and the available experimental data.

3 DIPOLE MOMENT CURVES

The DMC of the $X^2\Pi$ state was computed using MRCI/aug-ccpwCV5Z-DK (Dunning [1989;](#page-8-47) Raghavachari et al. [1989\)](#page-9-46) with the core-correlation and relativistic effects, the latter using the Douglas– Kroll–Hess method (Reiher [2006\)](#page-9-47) and including the Davidson correction (Langhoff & Davidson [1974\)](#page-8-48). The finite field method was used. For a discussion of the calculation of DMCs using the expectation and finite field methods see Lodi & Tennyson [\(2010\)](#page-8-49).

Figure 1. Potential energy curves of SiH used in the line list production. The $X^2\Pi$ and $A^2\Delta$ PECs have been refined, the $a^4\Sigma^-$ is ab initio and the $B^2\Sigma^-$ PEC is an artificial object used to improve the description of the -doubling in the *X*-state spectra (see Section 4).

Figure 2. Dipole moment curve calculated for the $X²$ Π state and the transition dipole moment for the $A^2\Delta - X^2\Pi$ state.

The dipole moment for the $X^2\Pi$ state and the transition dipole moment (TDM) between the $A^2\Delta$ and $X^2\Pi$ states are shown in Fig. [2.](#page-2-1) Our equilibrium value of the $X² \Pi$ DMC is 0.097 D. Previous theoretical estimates range between 0.076 D (Allen & Schaefer [1986\)](#page-8-50) and 0.173 D (Ajitha & Pal [1999\)](#page-8-15) with other estimates being 0.122 D (Larsson [1987;](#page-8-16) Park & Sun [1992\)](#page-9-11), 0.125 D (Mauricio et al. [1988\)](#page-8-51), 0.117 D (Pettersson & Langhoff [1986\)](#page-9-12), 0.140 D (Meyer & Rosmus [1975\)](#page-8-13) and 0.160 D (Huzinaga [1965\)](#page-8-52).

The ab initio TDM components μ_{α} between the $A^2\Delta$ and $X^2\Pi$ states can be defined as the matrix elements between the Cartesian $|\Delta_{\alpha}\rangle$ and $|\Pi_{\alpha}\rangle$ components of the corresponding electronic eigenfunctions (where $\alpha = x$, *y*, *z*), which is also MOLPRO's basis set convention. For the 'equilibrium' value of $\langle \Delta_z | \mu_y | \Pi_y \rangle$ (taken at $r = 1.52$ Å) we obtained 0.585 D. This matrix element is connected to the spherical tensor dipole representation by

$$
\langle \Lambda = 2|\mu_{+}| \Lambda = 1 \rangle = \sqrt{2} \langle \Delta_{x} | \mu_{y} | \Pi_{y} \rangle = 0.827 \,\mathrm{D},\tag{1}
$$

where $\mu_{+} = (-\mu_{x} + i\mu_{y})/\sqrt{2}$. Here $|1\rangle \equiv |^{2}\Pi\rangle$ and $|2\rangle \equiv |^{2}\Delta\rangle$ are eigenfunctions of the \hat{L}_z operator and also linear combinations of $| \Pi_x \rangle$, $| \Pi_y \rangle$ and $| \Delta_z \rangle$, $| \Delta_{xy} \rangle$, respectively. The tensorial representation of the dipole moment (μ_+) has been generally recommended (Whiting et al. [1980\)](#page-9-48). Larsson [\(1987\)](#page-8-16) obtained a CASSCF value of $\mu_+ = 0.706$ D using a 5-electrons in 9 orbitals (3 σ , 2π , 1 δ) active space (see table V of cited paper) and a basis set approximately equivalent in size to the cc-pVTZ one. It should be noted that the latter value better reproduces the observed $A - X$ lifetime (see discussion below). As suggested by Larsson [\(1987\)](#page-8-16), it is important to

Table 1. Expansion parameters of the dipole moment functions, $X - X \langle \Pi_x | \mu_z | \Pi_x \rangle$ and $X - A \langle \Delta_x | \mu_y | \Pi_y \rangle$, see equation (2). The units are Å and Debye.

Parameter	$X - X$	$X - A$
r_{ref}	1.5202	1.5202
β_2	0.216	0.460
β_4	0.059 795 80	0.01
р		8
d_0	0.097 003 6013	0.485 711 2345
d_1	-2.3766347830	0.5013535548
d_2	-1.3617433491	0.719 005 5685
d_3	0.114 661 2097	0.118 188 8386
d4	0.137 153 0465	-2.2686306105
d_5	0.243 634 3227	-2.2419525352
d6		2.372 986 8957
		$\mathbf{\Omega}$

include at least one set of (doubly degenerate) δ orbitals into CAS. We also found a strong variation of the $A - X$ transition dipole with respect to the active space used. Our final choice for the active space was a rather large 5-electrons in 15 orbitals complete active space comprising $(C_{2v}$ symmetry labels) 8 a_1 , 3 b_1 , 3 b_2 and 1 a_2 active orbitals; the 5 core orbitals (3 of a_1 symmetry and one for both b_1) and b_2 symmetries) were kept doubly occupied (i.e. excluded from the active space). CASSCF calculations used state-averaging over the lowest Σ^+ , Σ^- , Π and Δ states. Our choice of active space was also motivated by reason of numerical stability and convergence of the calculations (see also the discussion at the end of Section 5). The aug-pV(Q+d)Z basis set was used. Core-correlation and relativistic corrections had only marginal effects and were not taken into account for this property. In any case contributions from these effects tend to cancel (Tennyson [2014\)](#page-9-49).

In order to reduce the numerical noise when computing the linestrengths using the DUO program, we followed the recommendation by Medvedev et al. [\(2016\)](#page-8-53) and represented these two DMCs analytically. The following expansion with a damped-coordinate was employed to represent our dipole moment functions:

$$
\mu(r) = (1 - \xi) \sum_{n \ge 0} d_n z^n + d_\infty \xi,
$$
\n(2)

where ξ is the Šurkus variable (Šurkus, Rakauskas & Bolotin [1984\)](#page-9-50)

$$
\xi = \frac{r^p - r_{\text{ref}}^p}{r^p + r_{\text{ref}}^p},\tag{3}
$$

with p as a parameter, and r_{ref} as a reference position and z is given by

$$
z = (r - r_{\text{ref}}) e^{-\beta_2 (r - r_{\text{ref}})^2 - \beta_4 (r - r_{\text{ref}})^4}.
$$
 (4)

The expansion parameters d_n , d_∞ (the value of the dipole at $r \to$ ∞), β_2 and β_4 (damping factors) are collected in Table [1](#page-2-2) and are given in supplementary material as part of the DUO input file, while the functional form has been implemented into DUO.

The PECs shown in Fig. [1](#page-2-0) are those calculated in this work with the $X^2\Pi$ and $A^2\Delta$ states having been refined, $a^4\Sigma^-$ is ab initio and $B²\Sigma⁻$ is a reference curve used to improve the description of the Λ doubling (see discussion below). Qualitatively, in terms of general behaviour these compare favourably to Kalemos et al. [\(2002\)](#page-8-18) who used larger basis sets (aug-cc-pV5Z and aug-cc-pV6Z) to calculate a number of PECs and SOCs for the low-lying electronic states of SiH.

Study	Method	System			Wavenumber range (cm^{-1})
Klynning et al. (1979)	CTS	$A^2 \Lambda - X^2 \Pi$	$0.5 - 14.5$	(0, 0)	23 958 - 24 399
Davies et al. (1985)	DLAR	$X^2\Pi - X^2\Pi$	$0.5 - 9.5$	(1, 0)	1838-2094
Betrencourt et al. (1986)	FTS	$X^2\Pi - X^2\Pi$	$0.5 - 15.5$	(1, 0), (2, 1), (3, 2)	1704-2142
Ram et al. (1998)	FTS	$A^2 \Lambda - X^2 \Pi$	$1.5 - 18.5$	(0, 0), (1, 1)	23 644 - 24 461

Table 2. Summary of experimental data used for refining the ab initio *X*² Π PEC if SiH. CTS, Czerny–Turner spectrograph; DLAR, diode laser absorption spectroscopy; FTS, Fourier transform spectrometer.

Figure 3. The SO couplings (Cartesian representation as computed by MOLPRO). The $X - X$, $X - A$, $X - B$ SOCs were refined.

4 REFINEMENT

In order to refine our model for the $X^2\Pi$ and $A^2\Delta$ states the experimental frequencies were collected from the papers given in Table [2.](#page-3-0) These measurements span *J* up to 18.5 in the $(0 - 1)$, $(0 - 1)$ and $(1 – 1)$ bands, and *J* up to 10.5 in the $(2 – 0)$ and $(3 – 2)$ bands. Using the MARVEL program (Measured Active Rotational-Vibrational Energy Levels) (Furtenbacher, Császár & Tennyson [2007;](#page-8-54) Furten-bacher & Császár [2012\)](#page-8-55), 337 energy levels were determined from 894 transitions.

The ab initio PECs, SOCs and EAMCs were refined by fitting to these derived MARVEL energy levels, which were then complemented with the original experimental frequencies. We used the Extended Morse Oscillator analytical function to represent the PECs in the fits, which has the form

$$
V(r) = V_{e} + (A_{e} - V_{e}) \left[1 - \exp \left(- \sum_{k=0}^{N} B_{k} \xi^{k} (r - r_{e}) \right) \right]^{2}, \tag{5}
$$

where $D_e = A_e - V_e$ is the dissociation energy, r_e is an equilibrium distance of the PEC and ξ is the Surkus variable with $r_{\text{ref}} = r_{\text{e}}$. Note that *p* and *N* can have different values in the short ($r \le r_e$) and long $(r > r_e)$ regions, i.e. $p_s(N_s)$ and $p_l(N_l)$, respectively. The parameters are given as supplementary data together with the actual PECs for convenience.

During the fitting of the *X* and *A* states, various electronic couplings involving this electronic state were included. The refined coupling curves are shown in Figs 3 and 4 . To represent the $X - X$ and $A - A$ SOCs, the expansion in equation (2) was used. The non-diagonal $X - a$, $X - A$, $X - B$ SOCs as well as the EAMCs were morphed using the following expansion in terms of the Šurkus variable:

$$
F(\xi) = (1 - \xi) \sum_{k=0}^{N} B_k \xi^k + \xi B_{\infty}.
$$
 (6)

Figure 4. Electronic angular momentum coupling curves (Cartesian representation as in MOLPRO). Both the $X - A$ and $X - B$ EAMCs were refined.

The expansion parameters are given in the supplementary material as part of the DUO input file.

For Π states Λ -doubling is predominantly caused by the interaction with Σ -states (Brown & Merer [1979\)](#page-8-56). In the case of $X^2\Pi$ of SiH, the closest Σ state is $B^2\Sigma^-$. This interaction is still much weaker than the SO coupling, giving splittings of roughly 0.1 cm^{-1} for $J = 0.5$ (see Table [4\)](#page-5-0) with the two components having almost the same intensity (Verma [1965\)](#page-9-27). Theoretically, excited states such as the $A^2\Delta$ should also exhibit Δ -doubling, although because of the small splittings this has proved difficult to observe experimentally.

The EAM coupling between the $B^2\Sigma^-$ and $X^2\Pi$ states is shown in Fig. [4.](#page-3-2) Since the $B^2\Sigma^-$ PEC is weakly bound and DUO currently can only work with properly bound potential energy curves, we used a dummy curve with a shallow minimum for the $B^2\Sigma^-$ PEC as a reference object. This was sufficient to produce the correct -doubling effect for all *J* values considered. It should be noted that inclusion of an effective Λ -doubling curve is available in DUO (Yurchenko et al. [2016\)](#page-9-45) but did not produce correct *J*-dependence and therefore was discarded. In order to model the $X - B$ EAM coupling the Surkus-expansion in equation (6) was used.

As well as SO and EAMC, other empirical corrections such as Born–Oppenheimer breakdown (BOB; Le Roy [2007\)](#page-8-57) and spinrotation (SR; Kato [1993\)](#page-8-58) were used to provide greater accuracy, see Yurchenko et al. [\(2016\)](#page-9-45). The SR within the *X* state and the $X - X$ and $A - A$ SO couplings were represented by the form given by equation (2) with the remaining SOC, EAMC, SR and the BOB curves were represented by equation (6). The parameters obtained as well as all curves specifying our final model form part of the DUO inputs are provided in the supplementary material.

The dissociation energy is poorly constrained by the low-lying vibrational state included in the fit. Initially the D_0 value was set to the experimental estimate of 2.98 eV ($\pm 0.03 \text{ eV}$) from Berkowitz et al. [\(1987\)](#page-8-59) and then allowed to float to values near this value. Our final value for the dissociation energy D_0 for the *X* state is 3.020 eV, which corresponds to $D_e = 3.136 \text{ eV}$ and the zero-point energy of

Figure 5. Visual representation of difference in calculated and experimentally measured frequencies as a function of *J* for low-lying vibrational states of ²⁸SiH in its $X^2\Pi$ and $A^2\Delta$ electronic states.

941.23 cm⁻¹ ($v = 0, J = 0.5, +$). Our final dissociation energy D_0 of the *A* state is 0.84 eV ($D_e = 0.95 \text{ eV}$).

Fig. [5](#page-4-0) offers a visual comparison of the Obs.–Calc. residuals as a function of *J*, with the various vibrational and vibronic bands are indicated. Table [3](#page-4-1) presents a representative sample of data for low *J* for $v = 0$ showing the comparison and Table [4](#page-5-0) demonstrates the accuracy when the vibrational number v is varied. Four hundred *X* − *X* transition wavenumbers ($J' \le 16.5$) are reproduced with the root-mean-square (rms) error of 0.015 cm⁻¹, 51 pure rotational energies $(J < 11.5)$ collected from CDMS are reproduced with an rms error of 0.005 cm⁻¹, 494 rovibronic $A - X$ transition wavenumbers $(J \le 18.5)$ are reproduced with an rms error of 0.016 cm⁻¹. The accuracy of the pure rotational energies is comparable to that of the typical effective Hamiltonian approaches.

5 LINE LISTS

DUO was used to solve the fully coupled Schrödinger equation for the four lowest bound electronic states of SiH using our refined curves. The details of the DUO methodology used for building accurate, empirical line lists for diatomic molecules has been extensively discussed elsewhere (Patrascu et al. [2014;](#page-9-51) Patrascu, Tennyson & Yurchenko [2015;](#page-9-52) Lodi, Yurchenko & Tennyson [2015;](#page-8-60) Tennyson et al. [2016b;](#page-9-53) Yurchenko et al. [2016\)](#page-9-45). A grid-based sinc basis of 501 points spanning 1–4 Å was used, selecting the 40, 20, 40 and 10 lowest vibrational eigenfunctions of the $X^2\Pi$, $a^4\Sigma^-, A^2\Delta, B^2\Sigma^$ states, respectively.

The line list produced for 28 SiH contains 1 724 841 transitions and 11 785 states covering frequencies up to 31 340 cm−1. The line lists for the isotopologues ²⁹SiH, ³⁰SiH and ²⁸SiD were generated using the same methodology by simply changing the nuclear masses to the corresponding values (see Table [5\)](#page-5-1). For compactness and ease of use the line lists are separated into energy state and transitions files using the standard ExoMol format (Tennyson et al. [2016c\)](#page-9-38). Tables [6](#page-6-0) and [7](#page-6-1) show extracts from the states and transition files, respectively. The full line lists for all isotopologues considered can be downloaded from www.exomol.com and from the CDS data base.

As part o[f](#page-5-0) the ExoMol line lists, we now provide lifetimes and Landé *g*-factors for the states involved (Tennyson et al. [2016a\)](#page-9-54). The methodology used to compute them is detailed in Semenov,

Yurchenko & Tennyson [\(2017\)](#page-9-55). The lifetime of the *A* state ($v = 0$, $J \leq 11.5$) was measured in laser-induced fluorescence experiments by Bauer et al. [\(1984\)](#page-9-22) to be 534 \pm 23 ns and by Schmitt et al. (1984) to be 530 ± 2 ns, see also a review of other measurements by Bauer et al. [\(1984\)](#page-8-45). Our value is 400 ns for the $J' = 10.5 (0 - 0)$. Larsson [\(1987\)](#page-8-16) in his CASSCF calculations showed that the *X* − *A* TDMs depends strongly on the active space (see table V of cited paper) and that the closest agreement with the experimental lifetime was not for the largest one. Among Larsson [\(1987\)](#page-8-16)'s computed lifetimes the one closest to experiment is 484 ns (μ = 0.706 D) using a 5-electrons in 9 orbitals (3 σ , 2 π , 1 δ) active space. Using a similar 5-electrons in 10 orbitals active space comprising $5 a_1$, $2 b_1$, $2 b_1$ and $1 a_2$ orbitals we have obtained a similar value of the equilibrium dipole moment $(\mu_{+} = 0.706 \text{ D})$. However, we could not converge our calculations of the DMC with this choice of the active space for all the geometries considered; we managed to obtain convergence for all values of *r* between 0.7 and 5 Å using the large 5-electrons in 15 orbitals

\overline{J}	Parity	State	υ	Ω	Obs.	Calc.	Obs-Calc.
0.5	$^{+}$	X	1	0.5	1970.3041	1970.3130	-0.0089
0.5	$^{+}$	X	$\mathfrak{2}$	0.5	3869.6575	3869.6600	-0.0025
0.5	$+$	X	3	0.5	5698.7697	5698.7789	-0.0092
0.5		X	θ	-0.5	0.0859	0.0972	-0.0113
0.5		X	1	-0.5	1970.4103	1970.4108	-0.0005
0.5		X	\overline{c}	-0.5	3869.7405	3869.7585	-0.0180
0.5		X	3	-0.5	5698.8651	5698.8763	-0.0112
1.5	$^{+}$	X	θ	0.5	21.0376	21.0336	0.0040
1.5	$+$	X	$\overline{0}$	1.5	151.5592	151.5585	0.0007
1.5	$+$	X	1	0.5	1990.7716	1990.7826	-0.0110
1.5	$+$	X	1	1.5	2122.1435	2122.1638	-0.0203
1.5	$+$	X	$\overline{2}$	0.5	3889.5532	3889.5673	-0.0141
1.5	$+$	X	\overline{c}	1.5	4021.7593	4021.7763	-0.0170
1.5	$^{+}$	X	3	0.5	5718.0995	5718.1237	-0.0242
1.5	$^{+}$	\boldsymbol{A}	$\overline{0}$	1.5	24 268,0465	24 268.0614	-0.0149
1.5	$+$	\boldsymbol{A}	1	1.5	25 928.2856	25928.2823	0.0033
1.5		X	θ	-0.5	20.8283	20.8469	-0.0186
1.5		X	$\overline{0}$	-1.5	151.5420	151.5510	-0.0090
1.5		X	1	-0.5	1990.5945	1990.5943	0.0002
1.5		X	1	-1.5	2122.1346	2122.1566	-0.0220
1.5		X	\overline{c}	-0.5	3889.3696	3889.3775	-0.0079
1.5		X	$\mathfrak{2}$	-1.5	4021.7541	4021.7694	-0.0153
1.5		X	3	-0.5	5717.9357	5717.9355	0.0002
1.5		\boldsymbol{A}	$\overline{0}$	-1.5	24 268,0560	24 268.0614	-0.0054
1.5		\boldsymbol{A}	1	-1.5	25 928.2705	25928.2823	-0.0118

Table 4. Example of Obs.–Calc. residuals, in cm⁻¹, illustrating the vibrational accuracy of our refined model. Energies for ²⁸SiH are given relative to the $v = 0, J = 0.5, +$, $\Omega = 0.5$ level.

Table 5. Statistics for line lists for all four isotopologues of SiH.

	28 SiH	29 SiH	30 SiH	28 SiD
$J_{\rm max}$	82.5	82.5	82.5	113.5
v_{max} (cm ⁻¹)	31 337.3	31 337.3	31 337.3	31 337.3
$E'_{\rm max}$ (cm ⁻¹)	31 337.3	31 337.3	31 337.3	31 337.3
E''_{max} (cm ⁻¹)	31 337.3	31 337.3	31 337.3	31 337.3
Number of energies	11785	11796	11808	21 2 30
Number of lines	1724 841	1726584	1728 386	3520657

active space described at the end of Section 3 which, however, gave a slightly too high value of μ (0.789 D). We therefore decided to re-scale the (11,4,4,1) DMC by a factor $\sqrt{400/530} \approx 0.869$, thus bringing the lifetime of the $J' = 10.5 (0 - 0)$ level to 530 ns. It should be noted that the lifetime of the $v = 0$ [s](#page-5-1)tate gradually increases to 800 ns at $J = 32.5$.

6 EXAMPLES OF SPECTRA

The temperature at which spectra are simulated has a strong effect on the intensities produced. Some examples of absorption spectra at different temperatures are presented in Fig. [6.](#page-6-2) The structure of the strongest electronic bands is shown in Fig. [7.](#page-6-3) The $a^4\Sigma^- - X$ ${}^{2}\Pi$ band is dipole forbidden but can 'steal' intensity by interacting with the \overline{X} ² Π state[.](#page-6-0) Considering the relative contrast of this band, it would be interesting to see attempts of detecting this bands experimentally.

6.1 Comparisons of spectra

In order to test the quality of our theoretical line list, we present a number of comparisons with previous works. The CDMS catalogue contains a comparatively small number of transitions at low wavenumbers (Müller et al. 2005). For this reason, this work is only comparable in the $(0 - 0)$ band. The CDSM spectra include hyperfine splitting, which is ignored in our calculations. In order to provide a fair comparison of CDMS with our spectra, we have combined the CDMS hyperfine sub-structures into single lines. As can be seen from Fig. [8,](#page-6-4) the line positions are very similar, but the intensities differ by about a factor of 4. The latter is due to the difference in the dipole moments. The CDMS used the equilibrium ab initio dipole moment value 0.087 D by Meyer & Rosmus [\(1975\)](#page-8-13), while our value is μ_e is 0.097 D. With our state-of-the-art ab initio level of theory of MRCI/aug-cc-pwC5Z-DK we should provide a higher quality $X - X$ dipole moment. More importantly, our calculations also fully incorporate the effect of zero-point vibrational motion into the dipole matrix element by using a proper averaging over the vibrational wavefunctions, which is important for this band due to the sign change of the DMC close to the equilibrium bond length, see Fig. [2.](#page-2-1) As the result, for the vibrationally averaged dipole moment μ_0 is 0.0474 D, i.e. significantly smaller (about half the size) than the equilibrium dipole moment. The corresponding intensities are therefore four times weaker than those predicted by CDMS. Such an overestimate would lead, for example, to fourfold underestimates of the abundance of any observed SiH. We recommend that the CDMS data are updated.

Fig. [9](#page-7-0) shows a comparison with a spectrum generated using the SiH line list by Kurucz [\(2011\)](#page-8-44), where only the $A - X$ transitions were included. These and other spectra presented in this work were computed using our new program EXOCROSS (Yurchenko, Al-Refaie & Tennyson 2017). In the range where a comparison can be made (above 20 000 cm⁻¹), the main peak at 24 100 cm⁻¹ (0 − 0) agrees quite well, while Kurucz's $(1 - 0)$ and $(0 - 1)$ peaks on the rightand left-hand side from it are much weaker. The $(2 – 0)$ and $(0 – 2)$ bands agree well again, also for the absolute intensities. The disagreement could be due to our use of more modern experimental

Table 6. Extract from the states file of the 28 Si¹H line list.

\boldsymbol{n}	Energy (cm^{-1})	g_i		τ	g -factor	Parity	e/f	State	$\boldsymbol{\eta}$	Λ	Σ	Ω
	0.000000	4	0.5	inf	-0.000721	$^{+}$	ϵ	X2Pi	Ω		-0.5	0.5
	1970.313482	4	0.5	$8.4593E - 03$	-0.000723	$^{+}$	ϵ	X2Pi			-0.5	0.5
3	3869.660924	4	0.5	$4.5247E - 03$	-0.000724	$^{+}$	ϵ	X ₂ Pi	2		-0.5	0.5
	5698.780278	4	0.5	$3.2392E - 03$	-0.000725	$^{+}$	ℓ	X2Pi	3		-0.5	0.5
5	7458.470416	4	0.5	$2.6176E - 03$	-0.000724	$^{+}$	ϵ	X2Pi	4		-0.5	0.5
6	9148.745041	4	0.5	$2.2639E - 03$	-0.000723	$^{+}$	ϵ	X ₂ Pi	5		-0.5	0.5
	10768.764728	4	0.5	$2.0464E - 03$	-0.000721	$^{+}$	ℓ	X2Pi	6		-0.5	0.5
8	12316.884393	4	0.5	$1.9086E - 03$	-0.000717	$^{+}$	ϵ	X2Pi			-0.5	0.5
9	13749.386223	4	0.5	2.4180E+00	3.337 141	$^{+}$	ϵ	a4Sigma	Ω	Ω	0.5	0.5
10	13790.645552	4	0.5	$1.8223E - 03$	-0.000713	$+$	ϵ	X ₂ Pi	8		-0.5	0.5
11	15 187.015 658	4	0.5	$1.7723E - 03$	-0.000709	$+$	ℓ	X2Pi	9		-0.5	0.5
12	15859.208556	4	0.5	5.8285E-01	3.337 143	$^{+}$	ϵ	a4Sigma		Ω	0.5	0.5
13	16502.475508	4	0.5	$1.7500E - 03$	-0.000710	$^{+}$	ϵ	X2Pi	10		-0.5	0.5
14	17732.992112	4	0.5	1.7519E-03	-0.000707	$^{+}$	ϵ	X2Pi	11		-0.5	0.5
15	17783.585608	4	0.5	$2.8270E - 01$	3.337 142	$^{+}$	ϵ	a4Sigma	2	Ω	0.5	0.5
16	18874.096315	4	0.5	1.7780E-03	-0.000691	$^{+}$	ϵ	X2Pi	12		-0.5	0.5
17	19508.796071	4	0.5	$1.6795E - 01$	3.337 143	$^{+}$	ϵ	a4Sigma	3	Ω	0.5	0.5
18	19921.019747	4	0.5	1.8315E-03	-0.000671	$^{+}$	ϵ	X2Pi	13		-0.5	0.5
19	20868.688182	4	0.5	$1.9204E - 03$	-0.000643	$^{+}$	ϵ	X2Pi	14		-0.5	0.5
20	21 019 734 154	4	0.5	1.1238E-01	3.337 140	$^{+}$	ℓ	a4Sigma	4	Ω	0.5	0.5

n: State counting number; \tilde{E} : state energy in cm⁻¹; *g_i*: total statistical weight, equal to $g_{ns}(2J + 1)$; *J*: total angular momentum; τ: lifetime (s⁻¹); *g*-Landé factors; +/ −: total parity; *e*/*f*: rotationless parity; state: electronic state; v: state vibrational quantum number; Λ : projection of the electronic angular momentum; Σ : projection of the electronic spin; Ω : projection of the total angular momentum, $\Omega = \Lambda + \Sigma$.

Table 7. Extract from the transitions file of the ²⁸SiH line list.

f	Í	A_{fi} (s ⁻¹)	$\tilde{\nu}_{fi}$
1494	1882	$6.5269E - 07$	21.092521
9460	9670	3.4859E-08	21.142047
6251	6561	$1.1771E - 09$	21.143927
1582	1711	$1.0066E - 10$	21.144 601
575	706	$2.0164E - 11$	21.145.126
413	290	$1.4640E - 03$	21.153921
6969	7067	$7.8938E - 10$	21.161879
4731	5076	1.2289E-11	21.165 181
2255	2642	5.3380E-11	21.170838

f: Upper state counting number; *i*: lower state counting number; A_{fi} : Einstein-A coefficient in s⁻¹; \tilde{v}_{fi} : transition wavenumber in cm−1.

Figure 6. Comparison of the SiH absorption spectra at five different temperatures. The difference in intensity between 300 K and higher temperatures is most pronounced around 20 000 cm−1.

Figure 7. $T = 2000$ K absorption spectra of ²⁸SiH: $X - X$, $a - X$ and $A - X$ bands, where the $a - X$ band is dipole forbidden.

Figure 8. Comparison with available CDMS data for the $(0-0)$ transitions of 28SiH at 298 K. The difference in intensity is about 4.4 times due to the difference in the dipoles.

Figure 9. A comparison of absorption spectra produced from our line list with that of Kurucz [\(2011\)](#page-8-44) at 1000 K.

Figure 10. Upper display: experimental emission spectrum of SiH from a 100 mTorr pure silane radio frequency discharge produced by Stamou et al. [\(1997\)](#page-9-32) at 2000 K. Peak heights are normalized to the highest peak of the *R*¹ branch, $R_1(10.5)$. Lower display: theoretical emission spectrum of SiH at 2000 K. The peak intensities are similar to that from the experiment.

data, and to the use of higher level ab initio techniques, causing a discrepancy between the two results.

An additional measure of the accuracy of our line list is comparison to the experimental spectrum produced by Stamou et al. [\(1997\)](#page-9-32), which is presented in Fig. [10.](#page-7-1) The basic shapes of the spectra are the same, with the peak intensity occurring close to 4140 Å which is taken to be the *Q* branch. Furthermore, additional peaks at 4130 and 4104 Å are replicated, as is the shape below 4100 Å. Differences in intensities arise at longer wavelengths (although the shape remains consistent), most likely because of the non-thermal effects not properly considered here (we assumed a Boltzmann distribution and used a Gaussian line profile of 0.5 cm^{-1} HWHM).

6.2 Partition function

The partition function was calculated with DUO in steps of 1 K and was fitted to the following functional form (Vidler & Tennyson [2000\)](#page-9-57) given by

$$
\log_{10} Q(T) = \sum_{n=0}^{8} a_n (\log_{10} T)^n.
$$
 (7)

The fitted expansion parameters for 28 SiH are presented in Table [8.](#page-7-2) These parameters reproduce the temperature dependence of partition function of SiH with a relative rms error of 0.56 per cent; the fitting error increases to 1.8 per cent at $T = 5000$ K. This is still a

Table 8. Expansion coefficients for the partition function of 28SiH given by equation (7). Parameters for other isotopologues can be found in the supplementary material.

0.872 519 166 079 a ₀ 0.331 757 300 487 a ₁ -1.370132158260 a ₂ 2.297 149 082 100 a_3 -1.997501906630
a ₄
1.134 338 070 560 a ₅
-0.384224515822 a ₆
0.068 401 930 630 a ₇
-0.004881419966 a ₈

Figure 11. A comparison between the partition function produced from our line list and the theoretical partition function described by Sauval & Tatum [\(1984\)](#page-9-58). The latter values were multiplied by $g_{\text{ns}} = 2$.

very small error, and thus the fit can be said to reliably reproduce the partition function. The partition function and the expansion parameters for all four species are included into the supplementary materials.

Our partition function can be compared to that computed by Sauval & Tatum [\(1984\)](#page-9-58) and by Barklem & Collet [\(2016\)](#page-8-61). In order to directly compare with our calculated partition function, these partition functions need to be multiplied by the nuclear statistical weight $g_{ns} = 2$ (Tennyson et al. [2016c\)](#page-9-38). This is the so-called physics convention for the nuclear statistical weights, which ExoMol uses. Fig. [11](#page-7-3) shows this comparison. All three partition function agree well at high temperature. The values given by Barklem & Collet [\(2016\)](#page-8-61) at lower temperatures are too high. We believe that our function is sufficiently complete and more accurate at lower temperatures. At $T = 75$ K our partition function 35.278 compares well to the CDMS's pure rotational value at $T = 75$ K of 35.277; this agreement continues to very low temperature, where our partition function has the physically correct behaviour at $T \to 0$ K:

$$
Q(T) \approx 4(e^{-0 c_2/T} + e^{-0.097 c_2/T}) + \cdots.
$$
 (8)

where 4 is the total degeneracy, 0 and 0.097 cm⁻¹ are the two lowest term values at $J = 0.5$ and c_2 is the second radiation constant.

The nuclear statistical weights *g*ns used to produce the partition functions for 28 SiD, 29 SiH and 30 SiH (as well as their line lists) are 3, 4 and 2, respectively.

Figure 12. Lifetimes of the three lower electronic states of SiH.

7 CONCLUSION

Accurate and complete line lists for 28SiH and three minor isotopologues 29 SiH, 30 SiH and 28 SiD have been produced, displaying good agreement with existing theoretical and experimental data. The line lists are called SiGHTLY. The accuracy of the rotational line positions is comparable to the one obtainable with effective rotational Hamiltonians. In order to reproduce the Λ -doubling splitting, an interaction via an electronic angular momentum coupling curve with the $B^2\Sigma^-$ state was included using a very simple approximation to represent the PEC of the *B*-state.

The vibrationally averaged dipole moment μ_0 in the ground electronic state exhibit strong vibrational dependence and is about half in magnitude with respect to the equilibrium value. We suggest that the CDMS value of μ_0 should be updated.

The lifetimes computed for the $A - X$ rotational transitions $(J \le 13.5)$ using our best ab initio transition DMC (\approx 400 ns) are off by about a factor 1.3 from the corresponding experimental values (530 ns). Fig. [12](#page-8-62) presents an overview of our computed lifetimes. We use this experimental value to improve the ab initio transition DMC by scaling it by a factor $\sqrt{400/530} \approx 0.869$, which leads to identical lifetimes (within 0.1 ns). A higher level of ab initio theory is therefore needed in order to be confident that the dipole moments used are accurate. It would also be beneficial to have more experimental data sensitive to the dipole moment, e.g. intensities or lifetimes.

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REFERENCES

- Ajitha D., Pal S., 1999, Chem. Phys. Lett., 309, 457
- Allard F., Hauschildt P. H., 1995, ApJ, 445, 433
- Allen W. D., Schaefer H. F., 1986, Chem. Phys., 108, 243
- Babcock H. D., 1945, ApJ, 102, 154
- Baeck K. K., Lee Y. S., 1990, J. Chem. Phys., 93, 5775
- Barklem P. S., Collet R., 2016, A&A, 588, A96
- Barton E. J., Yurchenko S. N., Tennyson J., 2013, MNRAS, 434, 1469
- Bauer W., Becker K. H., Düren R., Hubrich C., Meuser R., 1984, Chem. Phys. Lett., 108, 560
- Berkowitz J., Greene J. P., Cho H., Ruscic B., 1987, J. Chem. Phys., 86, 1235
- Betrencourt M., Boudjaader D., Chollet P., Guelachvili G., Morillonchapey M., 1986, J. Chem. Phys., 84, 4121
- Bollmark P., Klynning L., Pages P., 1971, Phys. Scr., 3, 219 `
- Brown J. M., Merer A. J., 1979, J. Mol. Spectrosc., 74, 488
- Brown J. M., Robinson D., 1984, Mol. Phys., 51, 883
- Brown J. M., Watson J. K., 1977, J. Mol. Spectrosc., 65, 65
- Brown J. M., Curl R. F., Evenson K. M., 1984, J. Chem. Phys., 81, 2884
- Brown J. M., Curl R. F., Evenson K. M., 1985, ApJ, 292, 188
- Buenker R. J., 1986, Int. J. Quantum Chem., 29, 435
- Cade P. E., Huo W. M., 1967, J. Chem. Phys., 47, 614
- Chang Y. W., Sun H., 2003, Bull. Korean Chem. Soc., 24, 723
- Chang Y. W., Sun H., 2008, J. Phys. Chem. B, 112, 16135
- Cooper D. L., Richards W. G., 1981, J. Chem. Phys., 74, 96
- Davies P. B., Isaacs N. A., Johnson S. A., Russell D. K., 1985, J. Chem. Phys., 83, 2060
- Davis D. N., 1940, PASP, 52, 280
- de Almeida A. A., Singh P. D., 1978, Astrophys. Space Sci., 56, 415
- Douglas A. E., 1957, Can. J. Phys., 35, 71
- Drevillon B., Toulemonde M., 1985, J. Appl. Phys., 58, 535
- Dunning T. H., 1989, J. Chem. Phys., 90, 1007
- Freedman R. S., Irwin A. W., 1976, A&A, 53, 447
- Furtenbacher T., Császár A. G., 2012, J. Quant. Spectrosc. Radiat. Transfer, 113, 929
- Furtenbacher T., Császár A. G., Tennyson J., 2007, J. Mol. Spectrosc., 245, 115
- Grant D. J., Dixon D. A., 2009, J. Phys. Chem. A, 113, 3656
- Grevesse N., Sauval A. J., 1970, A&A, 9, 232
- Grevesse N., Sauval A. J., 1971, J. Quant. Spectrosc. Radiat. Transfer, 11, 65
- Herbst E., Millar T. J., Wlodek S., Bohme D. K., 1989, A&A, 222, 205
- Herzberg G., Lagerqvist A., McKenzie B. J., 1969, Can. J. Phys., 47, 1889
- Husain D., Norris P. E., 1979, Faraday Discuss., 67, 273
- Huzinaga S., 1965, J. Chem. Phys., 42, 1293
- Jackson C. V., 1930, Proc. R. Soc. A, 126, 373
- Jasinski J. M., Becerra R., Walsh R., 1995, Chem. Rev., 95, 1203
- Johnson R. D., Hudgens J. W., 1989, J. Phys. Chem., 93, 6268
- Jun L., Jun L., Zhenfu H., Zhenyu D., 2010, China Pet. Process. Petrochem. T., 12, 6
- Kalcher J., 1987, Chem. Phys., 118, 273
- Kalemos A., Mavridis A., Metropoulos A., 2002, J. Chem. Phys., 116, 6529
- Kampas F. J., Griffith R. W., 1981, J. Appl. Phys., 52, 1285
- Kato H., 1993, Bull. Chem. Soc. Jpn., 66, 3203
- Klynning L., Lindgren B., Sassenberg U., 1979, Phys. Scr., 20, 617
- Kurucz R. L., 2011, Can. J. Phys., 89, 417
- Lambert D. L., Mallia E. A., 1970, MNRAS, 148, 313
- Langhoff S. R., Davidson E. R., 1974, Int. J. Quantum Chem., 8, 61
- Larsson M., 1987, J. Chem. Phys., 86, 5018
- Le Roy R. J., 2007, LEVEL 8.0 A Computer Program for Solving the Radial Schrödinger Equation for Bound and Quasibound Levels. University of Waterloo Chemical Physics Research Report CP-663, <http://leroy.uwaterloo.ca/programs/>
- Leroy G., Sana M., Wilante C., Temsamani D. R., 1992, J. Mol. Struct. (THEOCHEM), 91, 369
- Lodi L., Tennyson J., 2010, J. Phys. B: At. Mol. Opt. Phys., 43, 133001
- Lodi L., Yurchenko S. N., Tennyson J., 2015, Mol. Phys., 113, 1559
- Lovas F. J., 1974, ApJ, 193, 265
- Matsuda A., Nakagawa K., Tanaka K., Matsumura M., Yamasaki S., Okushi H., Iizima S., 1980, J. Non-Cryst. Solids, 35, 183
- Mauricio M. O., Roos B. O., Sadlej A. J., Diercksen G. H. F., 1988, Chem. Phys., 119, 71
- McMillen D. F., Golden D. M., 1982, Annu. Rev. Phys. Chem., 33, 493
- Medvedev E. S., Meshkov V. V., Stolyarov A. V., Ushakov V. G., Gordon I. E., 2016, J. Mol. Spectrosc., 330, 36
- Merrill P. W., 1955, PASP, 67, 199
- Meyer W., Rosmus P., 1975, J. Chem. Phys., 63, 2356
- Moore-Sitterly C., 1966, Trans. Int. Astron. Union B, 12, 173
- Müller H. S. P., Schlöder F., Stutzki J., Winnewisser G., 2005, J. Mol. Struct. (THEOCHEM), 742, 215
- Nemoto M., Suzuki A., Nakamura H., Shibuya K., Obi K., 1989, Chem. Phys. Lett., 162, 467
- Owens A., Yurchenko S. N., Yachmenev A., Thiel W., Tennyson J., 2017, MNRAS, 471, 5025
- Park C., 1979, J. Quant. Spectrosc. Radiat. Transfer, 21, 373
- Park J. K., Sun H. S., 1992, Chem. Phys. Lett., 195, 469
- Patrascu A. T., Hill C., Tennyson J., Yurchenko S. N., 2014, J. Chem. Phys., 141, 144312
- Patrascu A. T., Tennyson J., Yurchenko S. N., 2015, MNRAS, 449, 3613
- Pearse R. W. B., 1933, Publ. Am. Astron. Soc., 7, 12
- Perrin J., Delafosse E., 1980, J. Phys. D: Appl. Phys., 13, 759
- Peterson K. A., Dunning T. H., 2002 J. Chem. Phys., 117, 10548
- Pettersson L. G. M., Langhoff S. R., 1986, Chem. Phys. Lett., 125, 429
- Prascher B. P., Lucente-Schultz R. M., Wilson A. K., 2009, Chem. Phys., 359, 1
- Qui-Xia L., Tao G., Yun-Guang Z., 2008, Chin. Phys. B, 17, 2040
- Raghavachari K., Trucks G. W., Pople J. A., Head-Gordon M., 1989, Chem. Phys. Lett., 157, 479
- Ram R. S., Engleman R., Bernath P. F., 1998, J. Mol. Spectrosc., 190, 341
- Ramakrishna Rao T., Lakshman S., 1971, Physica, 56, 322
- Reiher M., 2006, Theor. Chem. Acc., 116, 241
- Robertson R., Gallagher A., 1986, J. Appl. Phys., 59, 3402
- Rochester G. D., 1936, Z. Phys., 101, 769
- Rydbeck O. E. H., Elldér J., Irvine W. M., 1973, Nature, 246, 466
- Sauval A. J., 1969, Sol. Phys., 10, 319
- Sauval A. J., Tatum J. B., 1984, ApJS, 56, 193
- Sax A. F., Kalcher J., 1991, J. Phys. Chem., 95, 1768
- Schadee A., 1964, Bull. Astron. Inst. Neth., 17, 311
- Schmitt J. P. M., Gressier P., Krishnan M., de Rosny G., Perrin J., 1984, Chem. Phys., 84, 281
- Seebass W., Werner J., Urban W., Comben E. R., Brown J. M., 1987, Mol. Phys., 62, 161
- Semenov M., Yurchenko S. N., Tennyson J., 2017, J. Mol. Spectrosc., 330, 57
- Shi D.-H., Zhang J.-P., Sun J.-F., Zhu Z.-L., Yu B.-H., Liu Y.-F., 2008, J. Mol. Struct. (THEOCHEM), 851, 30
- Shi D., Li P., Zhu Z., Sun J., 2013, Spectra Chim. Acta A, 115, 259
- Singh P. D., Vanlandingham F. G., 1978, A&A, 66, 87
- Smith W. H., 1969, J. Chem. Phys., 51, 520
- Smith W., Liszt H., 1971, J. Quant. Spectrosc. Radiat. Transfer, 11, 45
- Song C., Gao T., Han H., Wan M., Yu Y., 2008, J. Mol. Struct. (THEOCHEM), 870, 65
- Stamou S., Mataras D., Rapakoulias D., 1997, Chem. Phys., 218, 57
- Stevens W. J., Krauss M., 1982, J. Chem. Phys., 76, 3834
- Šurkus A. A., Rakauskas R. J., Bolotin A. B., 1984, Chem. Phys. Lett., 105, 291
- Taniguchi M., Hirose M., Hamasaki T., Osaka Y., 1980, Appl. Phys. Lett., 37, 787
- Tennyson J., 2014, J. Mol. Spectrosc., 298, 1
- Tennyson J., Yurchenko S. N., 2012, MNRAS, 425, 21
- Tennyson J., Yurchenko S. N., 2017, Int. J. Quantum Chem., 117, 92
- Tennyson J., Hulme K., Naim O. K., Yurchenko S. N., 2016a, J. Phys. B: At. Mol. Opt. Phys., 49, 044002
- Tennyson J., Lodi L., McKemmish L. K., Yurchenko S. N., 2016b, J. Phys. B: At. Mol. Opt. Phys., 49, 102001
- Tennyson J. et al., 2016c, J. Mol. Spectrosc., 327, 73
- Turban G., Catherine Y., Grolleau B., 1980, Thin Solid Films, 67, 309
- Turban G., Catherine Y., Grolleau B., 1981, Thin Solid Films, 77, 287
- Turner J. L., Dalgarno A., 1977, ApJ, 213, 386
- Verma R. D., 1965, Can. J. Phys., 43, 2136
- Vidler M., Tennyson J., 2000, J. Chem. Phys., 113, 9766
- Visscher C., Lodders K., Fegley B., Jr, 2010, ApJ, 716, 1060
- Washida N., Matsumi Y., Hayashi T., Ibuki T., Hiraya A., Shobatake K., 1985, J. Chem. Phys., 83, 2769
- Weinreb S., Barrett A. H., Meeks M. L., Henry J. C., 1963, Nature, 200, 829
- Werner H. J., Knowles P. J., Lindh R., Manby F. R., Schütz M., 2010, MOLPRO, a package of ab initio programs
- Werner H.-J., Knowles P. J., Knizia G., Manby F. R., Schütz M., 2012, WIREs Comput. Mol. Sci., 2, 242
- Whiting E. E., Schadee A., Tatum J. B., Hougen J. T., Nicholls R. W., 1980, J. Mol. Spectrosc., 80, 249
- Wilson I. D. L., Richards W. G., 1975, Nature, 258, 133
- Wilson R. W., Penzias A. A., Jefferts K. B., Kutner M., Thaddeus P., 1971, ApJ, 167, L97
- Woon D. E., Dunning T. H., 1993, J. Chem. Phys., 98, 1358
- Yurchenko S. N., Lodi L., Tennyson J., Stolyarov A. V., 2016, Comput. Phys. Commun., 202, 262
- Yurchenko S. N., Al-Refaie A. F., Tennyson J., 2017, Comput. Phys. Commun., submitted

SUPPORTING INFORMATION

Supplementary data are available at *[MNRAS](https://academic.oup.com/mnras/article-lookup/doi/10.1093/mnras/stx2738#supplementary-data)* online.

Table 8. Expansion coefficients for the partition function of ²⁸SiH given by equation (7).

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