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Synthesis and analysis of adaptive Pd-integrated perovskite catalysts for effective NO_x-reduction

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Introduction:

The purpose of the present work is to develop a catalyst with improved NO_x-reduction capacity for lean-burn applications. The main idea is to chemically bond palladium ions into the perovskitic crystal structure to create oxygen vacancies and reduce the amount of the PGM. Palladium ions may reversibly diffuse in and out of the perovskite structure depending on the redox conditions [1]. This reversible movement may hinders sintering effects (i.e. particle growth) that otherwise cause deactivation of the catalyst during long-term service resulting in improved catalyst performance.

Results:



Evolution of phase(s) in LaFe $_{0.65}$ Co $_{0.3}$ Pd $_{0.05}$ O $_{3}$ as detected by XRD after calcination in air [2]. -Only the perovskite with orthorhombic phase was found

 PdÓ not found after calcination in air up to 700°C - After treatment at 900°C tetragonal PdO was observed

XPS analysis





LaFe_{0.65}Co_{0.35}O₃-Co_{0.55}O₅Co_{0.35}O₆₅Co_{0.35} v_{0.05}O₃ v_{0.05}O

- Three Pd-states are found in the perovskite surfaces: Pd³⁺. Pd²⁺ and Pd^o.

- Only Pd° is found on Pd-LaFe_{0.65}Co_{0.35}O₃ upon reduction at 200°C, - Pd° and Pd³⁺ are found even after reduction at 500°C on

 $LaFe_{0.65}Co_{0.3}Pd_{0.05}O_3,$ - Asymmetry of the XPS signals as indicator for Pd in the lattice?

Perovskite coating on cordierite substrates



Conclusions:

- ✤ Finer Pd-particles are obtained in the Pd-integrated perovskite LaFe_{0.65}Co_{0.3}Pd_{0.05}O₃ than in the Pd-supported perovskite Pd-LaFe_{0.65}Co_{0.35}O₃,
- TEM investigation suggests homogeneous distribution of Pd in the perovskite LaFe_{0.65}Co_{0.3}Pd_{0.05}O₃-700°C, * XPS study indicates that Pd diffuses partially out of the perovskite lattice upon reduction treatment as some lattice
- bonded Pd remains.
- Less production of N₂O of the perovskite based catalysts is related to the new Pd-sites (Pd-states, i.e. Pd³⁺), ◆ Creation of new Pd-sites (Pd-states, i.e. Pd³⁺) shift the NO₄-conversions to higher temperatures (C₃H₆-SCR of NO₄) of LaFe_{0.65}Co_{0.3}Pd_{0.05}O₃ in comparison to the NO_x-conversions of the Pd-supported perovskite Pd-LaFe_{0.65}Co_{0.35}O₃.

SEM and TEM study



LaFe,

La Ce Fe Pd O 250 300 200

perature ("C)

catalyst bed ten

SEM-micrograph of LaFe0.65Co0.3Pd0.05O3 reduced in 5 vol. % $\rm H_2$ in $\rm N_2$ at 600°C/3h

- Particle sizes 10 - 15 nm - Possible diffusion of Pd out of the perovskite lattice?

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SEM-micrograph of

- Pd particles supported on the perovskite surface. - Particle sizes 50 – 80 nm

H₂-SCR of NO_x



- Homogeneous distribution of Pd in the perovskite matrix.

- 5 nm Pd-particle.

- Pd on the perovskite surface - 20 nm Pd-particles.

C₃H₆-SCR of NO_x



NO₂-conversion (left) and N₂-selectivity (right) of the perovskites $\begin{array}{l} LaFe_{0,475}Co_{0,475}Pd_{0,5}O_{3}, La_{0,5}Cc_{0,475}Pd_{0,5}O_{3}-900^{\circ}C/3h} \ and \ 1 \ wt. \ \% \ Pt/SiO_{2}. \\ \hline Reaction \ conditions: \ 300 \ mg \ of \ catalyst, \ gas \ composition = \ 0.072 \ vol. \ \% \ NO + 5 \ vol. \ \% \ O_{2} + 1 \ vol. \ \% \ H_{2} + 7.2 \ vol. \ \% \ H_{2}O + 7.2 \ vol. \ \% \ CO_{2} + He, \ WF = \ 0.065 \ g.s.ml^{-1} \end{array}$ - Perovskite composition affects the NO.-conversions and N--selectivity of Pd

- The different Pd-states in the perovskites caused a positive effect to the NOx conversion and N₂-selectivity. Pt/SiO₂ catalyst produced higher amount of N₂O.

Peak C₃H₆-SCR of NO_x-conversions of the catalytic converter

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H. Tanaka, M. Misono, Current Opinion in Solid State and Materials Science 5 01) 381-387. [2] G.C. Mondragón Rodríguez, R. Ochrombel, B. Saruhan, Journal of European Ceramic Society 28 (2008) 2611-2616.



TEM-image of

Pd-LaFe_{0.65}Co_{0.35}O₃ reduced in 5 vol. % H₂ in

N₂ at 600°C/3h.

Reaction conditions: 75 mg of catalyst, gas composition = 0.05 vol.% NO + 0.05 vol.% C₃H₆ + 5 vol.% O₂ + Ar, W/F = 0.015 g.s. ml⁻¹.

- NO_x-conversion is affected by the state of Pd in the perovskites, - LaFe_{0.65}Co_{0.3}Pd_{0.05}O₃ display higher NO_x-reduction than Pd-LaFe_{0.65}Co_{0.35}O₃ specially at higher temperatures.

