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Desferrioxamine biosynthesis: diverse hydroxamate assembly by substrate-tolerant acyl transferase DesC

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Main Text

Summary

Hydroxamate groups play key roles in the biological function of diverse natural products. Important examples include trichostatin A, which inhibits histone deacetylases via coordination of the active site zinc(II) ion with a hydroxamate group, and the desferrioxamines, which use three hydroxamate groups to chelate ferric iron. Desferrioxamine biosynthesis in *Streptomyces* species involves the DesD-catalysed condensation of various *N*-acylated derivatives of *N*-hydroxycadaverine with two molecules of *N*-succinyl-*N*-hydroxycadaverine to form a range of linear and macrocyclic *tris*-hydroxamates. However, the mechanism for assembly of the various *N*-acyl-*N*-hydroxycadaverine substrates of DesD from *N*-hydroxycadaverine has until now been unclear. Here we show that the *desC* gene of *Streptomyces coelicolor* encodes the acyl transferase responsible for this process. DesC catalyses the *N*-acylation of *N*-hydroxycadaverine with acetyl, succinyl and myristoyl-CoA, accounting for the diverse array of desferrioxamines produced by *S. coelicolor*. The X-ray crystal structure of DesE, the ferrioxamine lipoprotein receptor, in complex with ferrioxamine B (which is derived from two units of *N*-succinyl-*N*-hydroxycadaverine and one of *N*-acetyl-*N*-hydroxycadaverine) was also determined. This shows that the acetyl group of ferrioxamine B is solvent exposed, suggesting that the corresponding acyl group in longer chain congeners can protrude from the binding pocket, providing insights into their likely function.

1. Introduction

A range of structurally-diverse microbial natural products contain one or more hydroxamate functional groups (Figure 1), which are employed as metal ion chelators to inhibit or enable important biological processes.¹ Prominent examples include trichostatin A (2), a histone deacetylase (HDAC) inhibitor with anticancer activity that is widely used in epigenetics research,² and desferrioxamine B (7), which is marketed as desferral for the treatment of iron overload and neuroblastoma in humans.^{3,4} Trichostatin A blocks histone deacetylation via chelation of its hydroxamate group to the active zinc(II) ion of HDACs,⁵ whereas desferrioxamine B uses its three hydroxamate groups to sequester ferric iron via formation of a hexadentate complex.⁶ Interestingly, desferrioxamine B (and other iron chelators) have been reported to enhance the anticancer activity of trichostatin A.⁷

Along with several other *bis*- and *tris*-hydroxamates, such as coelichelin (5), erythrochelin (1), vicibactin (3), aerobactin (6) and rhodochelin (4) (Figure 1), and other members of the desferrioxamine complex (Figure 2), desferrioxamine B (8) plays an important role in microbial iron uptake.⁸⁻¹¹ Iron is an essential nutrient for most forms of life because it plays a central role in several key metabolic processes. Although iron is the fourth most abundant element on Earth, its availability in aqueous environments is extremely low due to the insolubility of ferric oxide and hydroxide complexes.^{12,13} As a consequence, saprophytic microorganisms excrete small molecules known as siderophores that scavenge iron from the environment.¹⁴ The resulting ferric-siderophore complexes are taken up by the microbial cells using ATP-dependent transport mechanisms and the iron is released via reduction or degradation of the ligand.¹⁵ One well-studied example is the model antibiotic-producing Actinobacterium *Streptomyces coelicolor* A3(2), which biosynthesises and excretes coelichelin (5), desferrioxamine B (7) and desferrioxamine E (8).¹¹ Distinct cell surface-associated lipoprotein

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receptor components of ATP-binding cassette (ABC) transporters (CchF and DesE, respectively) are used by *S. coelicolor* to import the ferricoelichelin and ferrioxamine complexes.¹⁶

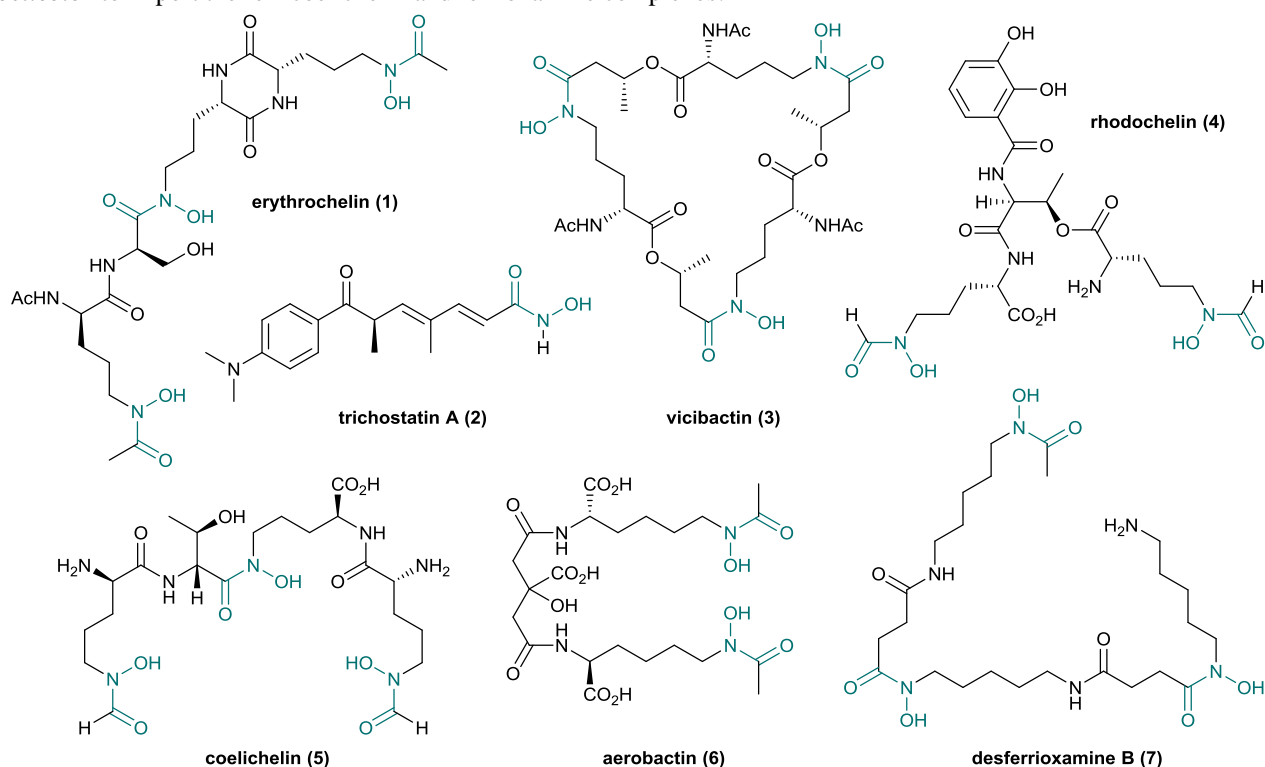


Figure 1: Chemical structures of some bacterial natural products containing hydroxamate groups (in grey).

Desferrioxamine E (**8**) and several other siderophores are also used by pathogenic microorganisms to scavenge iron from plant and animal hosts.¹⁷⁻¹⁹ The enzymes involved in the biosynthesis of such molecules thus offer potential as targets for the development of antibacterials. The mechanism for hydroxamate incorporation into several bacterial siderophores has been shown to involve initial oxidation of the amino group in the side chains of ornithine and lysine by flavin-dependent monooxygenases.^{8-10, 20-22} The resulting hydroxylamines undergo formylation by formyl tetrahydrofolate-dependent formyl transferases (e.g. in rhodochelin (**4**)¹⁰ and coelichelin (**5**)²⁰ biosynthesis), acylation by acyl-CoA-dependent acyl transferases (e.g. in aerobactin (**6**)²³ and vicibactin (**3**)⁹ biosynthesis), or condensation with amino acyl thioesters catalysed by nonribosomal peptide synthetase (NRPS) assembly lines.⁸ In contrast, trichostatin A (**2**) biosynthesis involves oxidation of the amido group in glutamine to the corresponding hydroxamate by a non-haem di-iron-dependent monooxygenase²⁴. An asparagine synthetase homologue then catalyses the hydrolysis of this hydroxamate, followed by ATP-dependent condensation of the resulting hydroxylamine with trichostatin acid (presumably via an acyl adenylate intermediate).

Desferrioxamine E (**8**) is assembled from three molecules of *N*-hydroxy-*N*-succinylcadaverine (HSC (**16**)) in *S. coelicolor* by the NRPS-independent siderophore (NIS) synthetase DesD (Figure 3).^{21, 25, 26} This process involves adenylation of the carboxyl group in one HSC unit followed by condensation with the amino group in another. The resulting homodimeric intermediate is adenylated and condensed with a third molecule of HSC to yield a homotrimer (desferrioxamine G₁ (**9**)). Desferrioxamine E is formed from this homotrimer via a further round of adenylation followed by macrolactamisation. DesD assembles desferrioxamine B (**7**) via condensation of the adenylated HSC homodimer with *N*-hydroxy-*N*-acetylcadaverine (HAC (**15**); Figure 3). Although *N*-hydroxycadaverine (**12**), arising from DesA-catalysed decarboxylation of L-lysine (**13**) followed by DesB-catalysed *N*-hydroxylation of the resulting cadaverine (**14**), is known to be an intermediate in desferrioxamine biosynthesis (Figure 3), the mechanism by which the key iron-chelating hydroxamate groups are incorporated into the desferrioxamines remains to be determined. The *desC* gene within the desferrioxamine biosynthetic gene cluster (which is highly conserved in many *Streptomyces* genomes) encodes a putative acyl-CoA dependent acyl transferase that has been proposed to catalyse the acylation of *N*-hydroxycadaverine with succinyl and acetyl-CoA to form HSC and HAC, respectively (Figure 3)²¹. However, this hypothesis remains experimentally untested and recent reports detailing the production of several desferrioxamine B analogues by *S. coelicolor*, in which the acetyl group is substituted with various nine to fifteen-carbon saturated acyl groups, further complicates the picture.^{27, 28} Presumably these amphiphilic desferrioxamines arise from the DesD-catalysed condensation of the HSC homodimer with the corresponding *N*-hydroxy-*N*-acyl-cadaverines, suggesting that DesC may be a remarkably substrate tolerant acyl transferase capable of forming the hydroxamate group in diverse desferrioxamine precursors. Moreover, the reason why *S. coelicolor* produces so many different desferrioxamine congeners remains unclear. Here we report experimental evidence demonstrating that this is indeed the case. We also report the X-ray crystal

structure of DesE (the desferrioxamine lipoprotein receptor) in complex with ferrioxamine B, leading us to propose that the amphiphilic desferrioxamines may function as membrane-bound iron shuttles in desferrioxamine-mediated iron uptake.

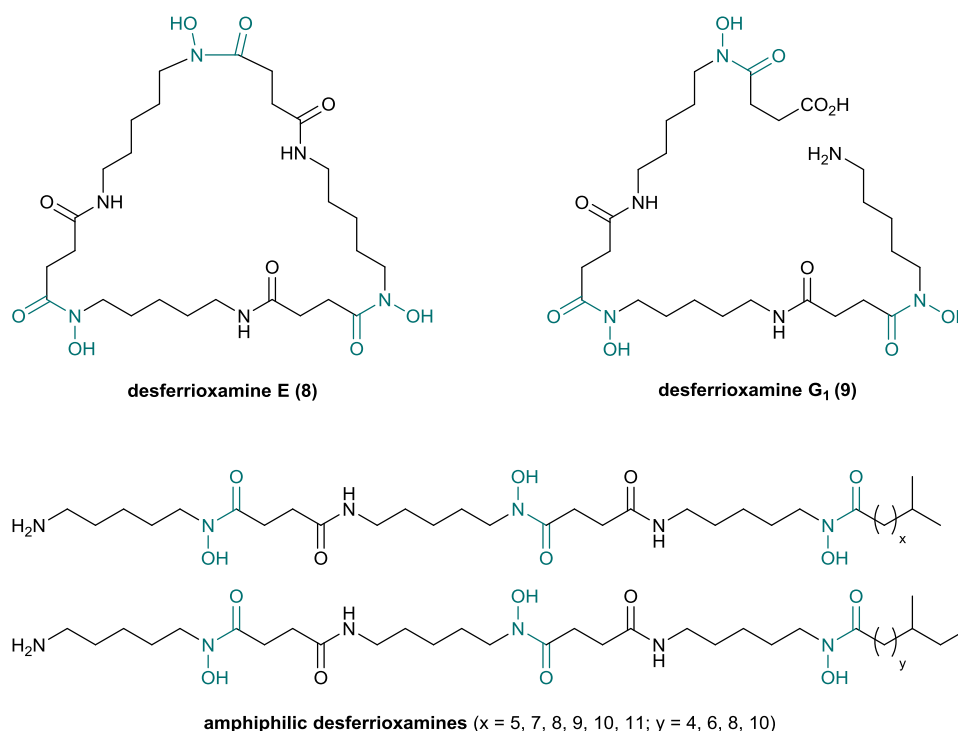


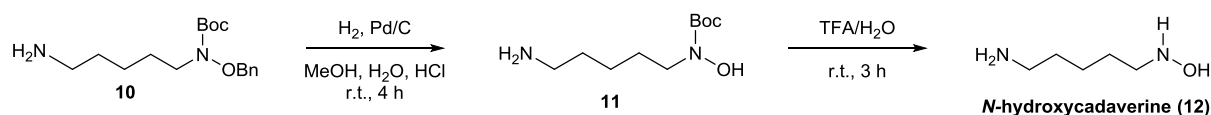
Figure 2: Chemical structures of desferrioxamine E and G₁, and the amphiphilic desferrioxamines reported to be produced, along with desferrioxamine B, by *S. coelicolor*.

2. Results and Discussion

(a) DesC catalyses *N*-acylation of *N*-hydroxycadaverine with diverse acyl-CoA thioesters

To investigate the catalytic properties of DesC, it was overproduced in *Escherichia coli* as a soluble *N*-terminal hexahistidine fusion and purified to homogeneity using nickel affinity chromatography (see supplementary material for experimental details). The identity of the purified protein was confirmed by peptide mass fingerprinting and gel filtration chromatography showed that it was monomeric (see supplementary material).

N-hydroxycadaverine (**12**) was synthesised from *N*-benzyloxy-*N*-BOC-cadaverine (**10**)²⁶ via hydrogenolytic cleavage of the benzyl ether, followed by removal of the BOC group under acidic conditions (Scheme 1; see supplementary material for experimental details). To establish whether DesC is able to catalyse the assembly of HSC (**16**), the core hydroxamate building block common to all desferrioxamines, the purified recombinant enzyme was incubated with *N*-hydroxycadaverine and succinyl-CoA for 10 minutes at 37°C. The reaction was stopped by addition of trichloroacetic acid and the mixture was analysed by LC-MS. A prominent species with $m/z = 219$, corresponding to the $[M+H]^+$ ion for HSC, was observed in these analyses (Figure 4). This product was confirmed as HSC by LC-MS/MS comparisons with an authentic synthetic standard, as described previously.²⁶ The reaction was monitored using a continuous spectroscopic assay employing 5,5'-dithiobis(2-nitrobenzoic acid) (DTNB) to detect the production of coenzyme A (see supplementary material). A small quantity of HSC was observed in control reactions from which the enzyme had been omitted (Figure 4). This is unsurprising given the high electrophilicity and nucleophilicity of the thioester group in succinyl-CoA and the hydroxylamino group in *N*-hydroxycadaverine, respectively. Negligible production of coenzyme A was observed in the control reaction using the DTNB assay (see supplementary material). Incubation of DesC with *N*-hydroxycadaverine and acetyl-CoA resulted in a product with $m/z = 161$, corresponding to the $[M+H]^+$ ion for HAC (**15**), in LC-MS analyses (Figure 4). As before, this was confirmed as HAC by comparison with an authentic standard,²⁶ coenzyme A was confirmed as the by-product of the reaction using the DTNB assay (see supplementary material) and small quantities of HAC were observed in the no enzyme control.



Scheme 1: Synthesis of *N*-hydroxycadaverine from *N*-benzyloxy-*N*-BOC-cadaverine.

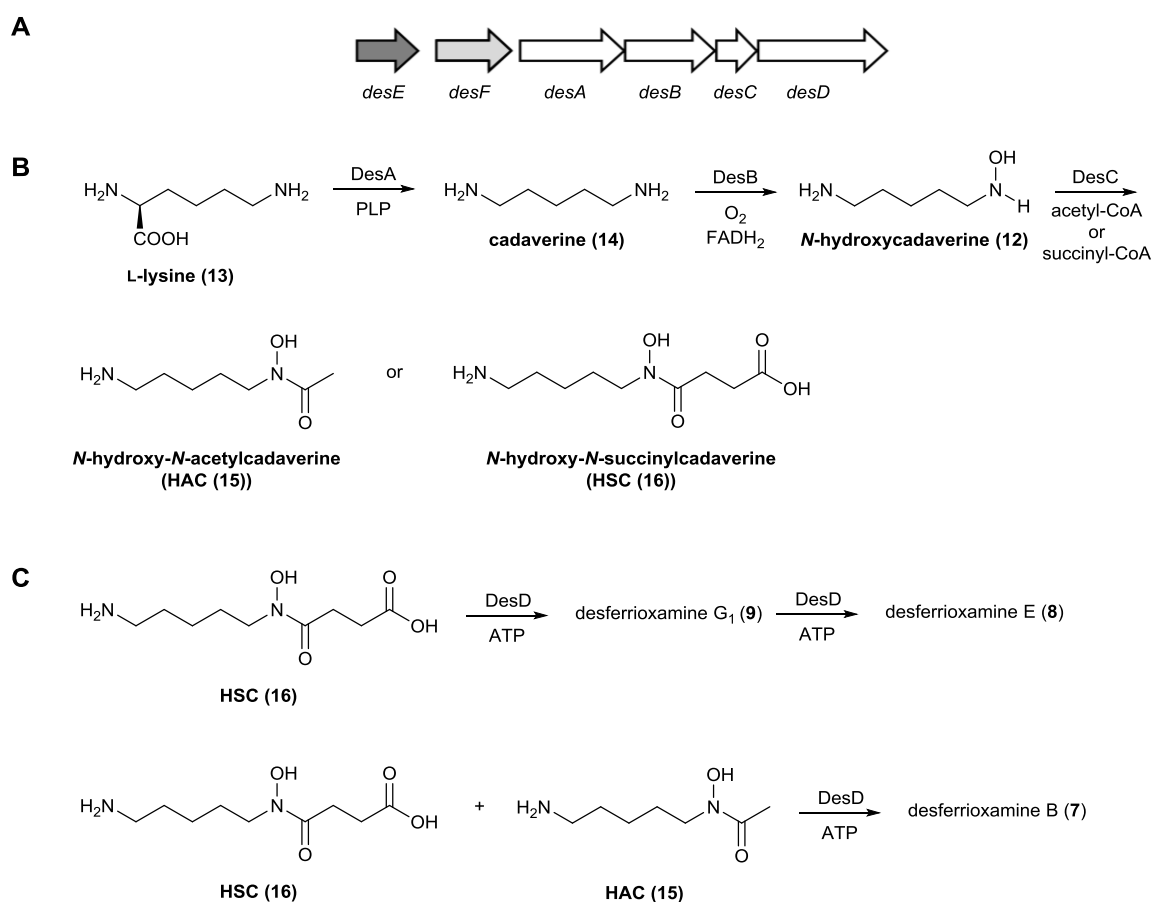
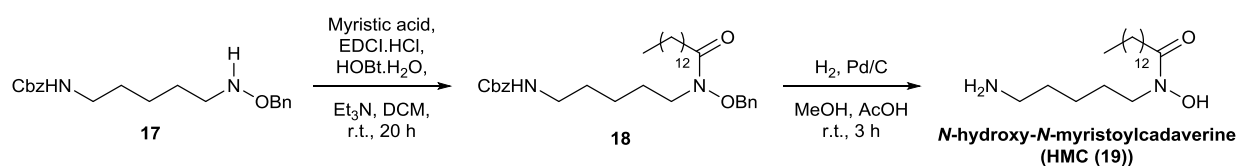


Figure 3: (A) Organisation of the desferrioxamine biosynthetic gene cluster in *S. coelicolor*. Genes encoding the four desferrioxamine biosynthetic enzymes (DesA-D) are white. The *desE* gene (dark grey) encodes a cell surface-associated lipoprotein receptor component of an ABC transporter involved in ferrioxamine uptake and *desF* (light grey) encodes a putative ferrioxamine reductase. (B) Previously proposed pathway for the assembly of HSC and HAC by the DesA, DesB and DesC enzymes. (C) DesD-catalysed assembly of desferrioxamine E, via desferrioxamine G₁, from three molecules of HSC, and the assembly of desferrioxamine B from two units of HSC and one of HAC.

Having established that DesC is able to biosynthesise not only the HSC (16) building block used by DesD for the assembly of desferrioxamine E (8), but also the HAC (15) unit used, along with HSC, for the assembly of desferrioxamine B (7),²⁶ we next investigated whether DesC is also able to produce medium chain acyl hydroxamates similar to those incorporated into the amphiphilic desferrioxamines. Thus, DesC was incubated with *N*-hydroxycadaverine (12) and myristoyl-CoA and the reaction mixture was analysed by LC-MS. A prominent species with $m/z = 329$, corresponding to the $[M+H]^+$ ion for *N*-hydroxy-*N*-myristoylcadaverine (HMC (19)), was detected (Figure 4). An authentic standard of HMC was synthesised via EDCI-mediated coupling of *N*-benzyloxy-*N'*-(benzyloxycarbonyl)cadaverine (17)²⁶ with myristic acid, followed by deprotection with H₂ over Pd/C in MeOH/MeCO₂H (Scheme 2; see supplementary material for experimental details). The product of the enzymatic reaction and the authentic standard of HMC had identical retention times and fragmentation patterns in LC-MS/MS analyses (see supplementary material). As observed with acetyl and succinyl-CoA (see above), a small amount of spontaneously generated HMC was present in the negative control (Figure 4).



Scheme 2: Synthesis of an authentic standard of HMC from *N*-benzyloxy-*N'*-(benzyloxycarbonyl)cadaverine.

Our data show that DesC is the key hydroxamate-forming enzyme in the desferrioxamine biosynthetic pathway. It not only catalyses the acylation of *N*-hydroxycadaverine with succinyl-CoA to form HSC, the precursor of desferrioxamines G₁, and E, but also the acylation of *N*-hydroxycadaverine with acetyl-CoA to form HAC, which is used along with HSC for the assembly of desferrioxamine B. Remarkably, DesC is also able to catalyse the acylation of *N*-hydroxycadaverine

with myristoyl-CoA (Figure 5), suggesting that it assembles the various *N*-hydroxy-*N*-acylcadaverines incorporated into the amphiphilic desferrioxamines. The broad substrate tolerance exhibited by DesC is atypical for acyl transferases involved in hydroxamate biosynthesis (e.g. IucB, which shows a high degree of specificity towards acetyl-CoA)²³.

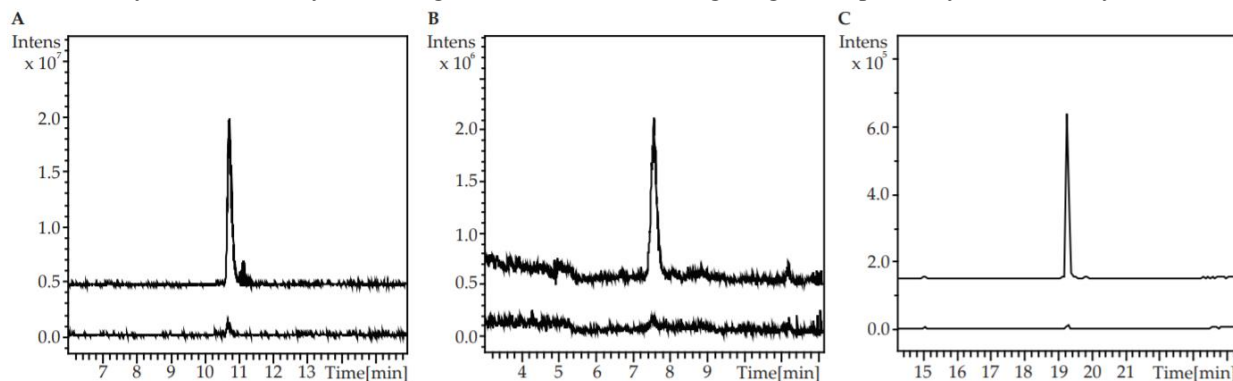


Figure 4: Extracted ion chromatograms (EICs) from LC-MS analyses of DesC-catalysed reactions (top traces) and control reactions from which the enzyme was omitted (bottom traces). (A) EICs at $m/z = 219$ (corresponding to $[M+H]^+$ for HSC (**16**)); (B) EICs at $m/z = 161$ (corresponding to $[M+H]^+$ for HAC (**15**)); (C) EICs at $m/z = 329$ (corresponding to $[M+H]^+$ for HMC (**19**)).

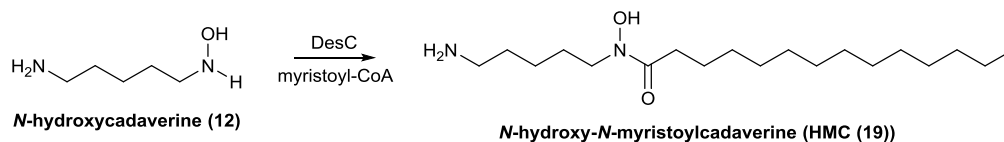


Figure 5: DesC-catalysed acylation of *N*-hydroxycadaverine with myristoyl-CoA to form HMC, suggesting that DesC can assemble the various *N*-hydroxy-*N*-acylcadaverines incorporated into the amphiphilic desferrioxamines.

(b) X-ray crystal structure of DesE-ferrioxamine B complex

We previously reported the X-ray crystal structure of *S. coelicolor* DesE,²⁹ an extracellular lipoprotein that binds ferrioxamines E/B (the ferric complexes of the corresponding desferrioxamines),¹⁶ initiating the uptake of these ferric-siderophore complexes via an ABC transporter.¹¹ To gain further insight into the mechanism of ferrioxamine recognition by DesE, we determined the crystal structure of the DesE-ferrioxamine B complex to 1.96 Å resolution (Figure 6; see supplementary material for experimental details). The conformation of DesE in the *apo* and ferrioxamine B-bound forms is essentially identical (see supplementary information). Visual inspection of the structure shows that the ferric ion is coordinated by the six oxygen atoms of the three hydroxamate groups in desferrioxamine B, as expected, with Fe-O distances between 1.9 and 2.0 Å. The substrate binding site of DesE is lined with the side chains of Trp122, Trp150, Trp270, Phe143, Phe271, Pro322 and Tyr236, which make multiple van der Waals contacts with hydrophobic regions of ferrioxamine B. The only direct polar contact between DesE and ferrioxamine B is mediated by the guanidinium group of Arg292, which is positioned within hydrogen bonding distance of one of the hydroxamate O atoms. Five ordered water molecules mediate indirect polar contacts between Ser243, Ser294, Ser151 and Pro322 of DesE with three of the oxygen atoms in ferrioxamine B. The *N*-acetyl group of ferrioxamine B points out the binding cleft. This indicates that DesE can also bind ferric complexes of the amphiphilic desferrioxamines, suggesting they may act as membrane-embedded shuttles that receive ferric ions scavenged from the environment by soluble siderophores, such as desferrioxamines B/E (**7/8**) and coelichelin (**5**).

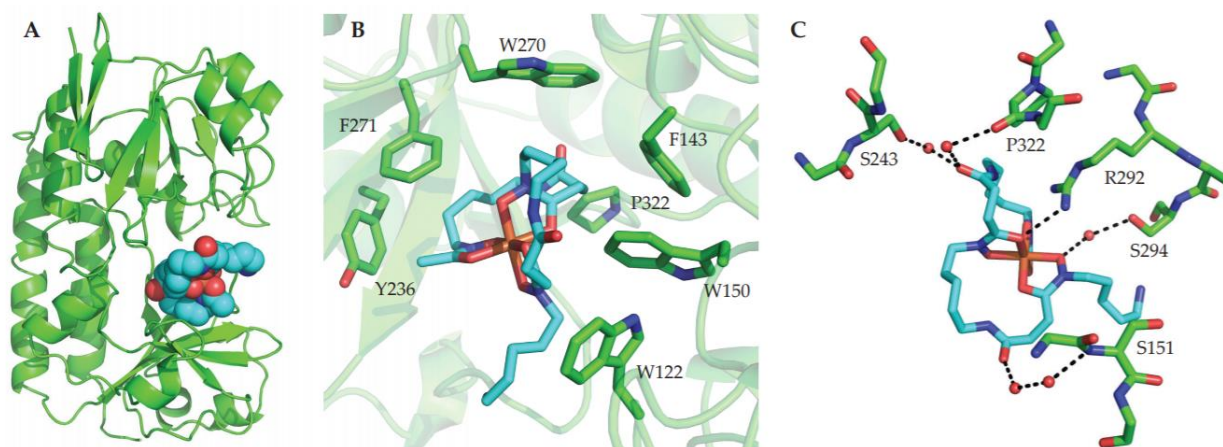


Figure 6: X-ray crystal structure of DesE (green) bound to ferrioxamine B (cyan). (A) Overall structure of the complex; (B) the ferrioxamine binding site, highlighting the hydrophobic residues lining it; (C) polar contacts between DesE and ferrioxamine B.

3. Conclusion

In this work we have shown that DesC is a remarkably substrate tolerant acyl transferase that catalyses the key hydroxamate-forming step in desferrioxamine biosynthesis. DesC is not only able to catalyse the acylation of *N*-hydroxycadaverine with succinyl and acetyl-CoA to form HSC and HAC, known intermediates in the biosynthesis of desferrioxamines E and B, but also with myristoyl-CoA to form HMC. This provides a rationale for the recently reported production of amphiphilic desferrioxamines by *S. coelicolor*. While the biological function of these unusual *tris*-hydroxamates remains unclear, the X-ray crystal structure of the DesE-ferrioxamine B complex suggests that ferric complexes of the amphiphilic desferrioxamines are likely to be recognised and imported by the *S. coelicolor* ferrioxamine uptake system. The nine to fifteen-carbon acyl chains in these molecules likely targets them to the extracytoplasmic membrane. It is therefore conceivable that ferric iron scavenged from the environment by desferrioxamines E/B and other soluble siderophores is transferred to the membrane-associated ferrioxamines prior to DesE-initiated uptake. Further experiments will be required to test this hypothesis.

Desferrioxamine B is used in the clinic as a treatment for iron overload. It will therefore be interesting to investigate whether the broad substrate tolerance of DesC can be harnessed for chemoenzymatic synthesis of novel desferrioxamine B analogues with improved therapeutic properties. DesC may also prove to be a useful tool in the development of biocatalytic approaches for the synthesis of hydroxamate-based HDAC inhibitors.

Additional Information

Acknowledgements

Dr Lijiang Song, Dr Ivan Prokes and Mr Philip Aston are thanked for assistance in obtaining spectroscopic data.

Data Accessibility

Supporting data and experimental procedures are included in the Supplementary Material. The coordinates of the DesE-ferrioxamine B complex have been deposited in the Protein Data Bank (Accession No. 6ENK).

Authors' Contributions

G.L.C. conceived the study and G.L.C., N.K., J.L.R., S.A.M. and J.H.N. designed the experiments. N.K., J.L.R. and S.A.M. acquired the data, and all authors contributed to its analysis and interpretation. G.L.C., L.M.A, N.K., J.L.R. and S.A.M. drafted the article and all authors approved the content.

Competing Interests

The authors have no competing interests.

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Supplementary material

Experimental procedures and additional data are provided in the supplementary material.