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# ELECTRON-IMPACT EXCITATION OF Cr ii: A THEORETICAL CALCULATION OF EFFECTIVE COLLISION STRENGTHS FOR OPTICALLY ALLOWED TRANSITIONS 

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#### Abstract

In this paper, we present electron-impact excitation collision strengths and Maxwellian averaged effective collision strengths for the complicated iron-peak ion Cr II. We consider specifically the allowed lines for transitions from the $3 d^{5}$ and $3 d^{4} 4 s$ even parity configuration states to the $3 d^{4} 4 p$ odd parity configuration levels. The parallel suite of $R$-Matrix packages, RMATRX II, which have recently been extended to allow for the inclusion of relativistic effects, were used to compute the collision cross sections. A total of $108 \mathrm{LS} \pi / 280 J \pi$ levels from the basis configurations $3 d^{5}, 3 d^{4} 4 s$, and $3 d^{4} 4 p$ were included in the wavefunction representation of the target including all doublet, quartet, and sextet terms. Configuration interaction and correlation effects were carefully considered by the inclusion of seven more configurations and a pseudo-corrector $\overline{4 d}$ type orbital. The 10 configurations incorporated into the Cr II model thus listed are $3 d^{5}, 3 d^{4} 4 s, 3 d^{4} 4 p, 3 d^{3} 4 s^{2}, 3 d^{3} 4 p^{2}, 3 d^{3} 4 s 4 p, 3 d^{4} \overline{4 d}, 3 d^{3} 4 s \overline{4 d}, 3 d^{3} 4 p \overline{4 d}$, and $3 d^{3} \overline{4 d}^{2}$, constituting the largest $\mathrm{Cr}_{\text {II }}$ target model considered to date in a scattering calculation. The Maxwellian averaged effective collision strengths are computed for a wide range of electron temperatures $2000-100,000 \mathrm{~K}$ which are astrophysically significant. Care has been taken to ensure that the partial wave contributions to the collision strengths for these allowed lines have converged with "top-up" from the Burgess-Tully sum rule incorporated. Comparisons are made with the results of Bautista et al. and significant differences are found for some of the optically allowed lines considered.


Key words: atomic data - atomic processes - methods: numerical - plasmas - scattering
Online-only material: color figures, machine-readable tables

## 1. INTRODUCTION

Cr II is a member of the astrophysically significant, yet notoriously difficult to model, iron-peak elements. These open $d$-shell ions, which range from scandium $(Z=21)$ to copper ( $Z=29$ ) in the periodic table, occur in abundance in many different astrophysical bodies and indeed, emission and absorption lines of these elements have been identified in the spectra of a wide variety of astronomical objects at all wavelengths from the infrared to the ultraviolet. Modeling these ions, however, is a difficult task due to their complicated structures caused by the presence of the open $d$-shells. Hundreds of fine-structure states need to be included to properly represent the Cr II ion leading to thousands of coupled channels in the scattering calculation.

Atomic data for Cr II are well known to be of critical importance in the analysis of a broad range of stellar and nebular spectra. For example, Iijima \& Nakanishi (2008) reported the observation of prominent emission lines of Cr II, as well as many other iron-peak elements, during the early decline stage of the peculiar, explosive object Nova (V445) Puppis 2000. In addition, Luftinger et al. (2010) used Doppler and Magnetic Doppler imaging techniques to derive the magnetic field geometry of the CoRoT CP2 target star HD 50773. Surface abundance distributions were derived for $\mathrm{Mg}, \mathrm{Si}, \mathrm{Ca}$, $\mathrm{Ti}, \mathrm{Cr}, \mathrm{Fe}, \mathrm{Ni}, \mathrm{Y}$, and Cu , with seven lines of Cr II used for abundance analysis in the co-added spectrum of HD 50773. Finally, Bergemann \& Cescutti (2010) have recently applied theoretical atomic data consisting of 114 levels for Cr I and 225 levels for Cr II when investigating non-local thermodynamic equilibrium abundances of Cr in metal-poor stars. This work was of particular interest in looking at star formation theory and the early evolution stages of a star.

Up until very recently there was a distinct paucity of atomic data available in the literature for the electron-impact excitation of Cr II. The earliest sophisticated calculation performed for this ion was reported by Bautista et al. (2009) where electronimpact collision strengths for transitions among the lowest 162 fine-structure levels of Cr II were evaluated but not published. Wasson et al. (2010) subsequently extended this calculation and computed the fine-structure collisional data for all forbidden transitions among the lowest 280 levels of Cr II arising from the basic configurations $3 d^{5}, 3 d^{4} 4 s$, and $3 d^{4} 4 p$. This latter calculation corresponded to a substantial 1932 coupled channel problem. The authors, however, concentrated on the fine-structure lines among the lowest 74 fine-structure levels corresponding to the $3 d^{5}$ and $3 d^{4} 4 s$ even parity configuration states. These transitions converge much more quickly than the allowed lines and the collision strength contributions from each of the $J \pi$ partial waves were found to have well converged before a total angular momentum value of $2 J=14$ had been reached. The same is not true for the allowed transitions considered in the present publications and the contributions from the higher partial waves need to be taken into account. In this paper we therefore concentrate on the optically allowed transitions from the $3 d^{5}$ and $3 d^{4} 4 s$ even parity levels to the $3 d^{4} 4 p$ odd parity states. The collision strengths are evaluated for a very fine mesh of incident electron energies and the contributions to the collision strengths from the higher partial waves have fully converged.

## 2. THE TARGET MODEL

The theoretical $\mathrm{Cr}_{\text {II }}$ model adopted in the present calculation has been described in detail by Wasson et al. (2010) and will only be summarized here. A total of $108 \mathrm{LS} / 280 j j$ levels formed from the basis configurations $3 d^{5}, 3 d^{4} 4 s$, and $3 d^{4} 4 p$

Table 1
Energy Levels for All Fine-structure Target States of Cr II Considered in the Present Work

| Index | Config. | LS | $J$ | Energy | Index | Config. | LS | $J$ | Energy |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $3 d^{5}$ | ${ }^{6} S^{e}$ | 5/2 | 0.0000000 | 65 | $3 d^{4}(a 1 G) 4 s$ | ${ }^{2} G^{e}$ | 9/2 | 0.3629062 |
| 2 | $3 d^{4}(5 D) 4 s$ | ${ }^{6} D^{e}$ | 1/2 | 0.1090039 | 66 | $3 d^{5}$ | ${ }^{2} F^{e}$ | 7/2 | 0.3633863 |
| 3 | $3 d^{4}(5 D) 4 s$ | ${ }^{6} D^{e}$ | 3/2 | 0.1096489 | 67 | $3 d^{4}(1 I) 4 s$ | ${ }^{2} I^{e}$ | 13/2 | 0.3663484 |
| 4 | $3 d^{4}(5 D) 4 s$ | ${ }^{6} D^{e}$ | 5/2 | 0.1106990 | 68 | $3 d^{4}(1 I) 4 s$ | ${ }^{2} I^{e}$ | 11/2 | 0.3665872 |
| 5 | $3 d^{4}(5 D) 4 s$ | ${ }^{6} D^{e}$ | 7/2 | 0.1121209 | 69 | $3 d^{4}(1 S) 4 s$ | ${ }^{2} S^{e}$ | 1/2 | 0.3682891 |
| 6 | $3 d^{4}(5 D) 4 s$ | ${ }^{6} D^{e}$ | 9/2 | 0.1138759 | 70 | $3 d^{4}(3 D) 4 s$ | ${ }^{2} D^{e}$ | 5/2 | 0.3909149 |
| 7 | $3 d^{4}(5 D) 4 s$ | ${ }^{4} D^{e}$ | 1/2 | 0.1779544 | 71 | $3 d^{4}(3 D) 4 s$ | ${ }^{2} D^{e}$ | 3/2 | 0.3917226 |
| 8 | $3 d^{4}(5 D) 4 s$ | ${ }^{4} D^{e}$ | 3/2 | 0.1788922 | 72 | $3 d^{5}$ | ${ }^{2} S^{e}$ | 1/2 | 0.4037556 |
| 9 | $3 d^{4}(5 D) 4 s$ | ${ }^{4} D^{e}$ | 5/2 | 0.1804114 | 73 | $3 d^{4}(1 D) 4 s$ | ${ }^{2} D^{e}$ | 3/2 | 0.4161696 |
| 10 | $3 d^{4}(5 D) 4 s$ | ${ }^{4} D^{e}$ | 7/2 | 0.1824721 | 74 | $3 d^{4}(1 D) 4 s$ | ${ }^{2} D^{e}$ | 5/2 | 0.4167274 |
| 11 | $3 d^{5}$ | ${ }^{4} G^{e}$ | 5/2 | 0.1869195 | 75 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} F^{o}$ | 1/2 | 0.4266858 |
| 12 | $3 d^{5}$ | ${ }^{4} G^{e}$ | 11/2 | 0.1869199 | 76 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} F^{o}$ | 3/2 | 0.4274310 |
| 13 | $3 d^{5}$ | ${ }^{4} G^{e}$ | 7/2 | 0.1869721 | 77 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} F^{o}$ | 5/2 | 0.4286629 |
| 14 | $3 d^{5}$ | ${ }^{4} G^{e}$ | 9/2 | 0.1869857 | 78 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} F^{o}$ | 7/2 | 0.4303659 |
| 15 | $3 d^{5}$ | ${ }^{4} P^{e}$ | 5/2 | 0.1988613 | 79 | $3 d^{5}$ | ${ }^{2} D^{e}$ | 5/2 | 0.4315251 |
| 16 | $3 d^{5}$ | ${ }^{4} P^{e}$ | 1/2 | 0.1988733 | 80 | $3 d^{5}$ | ${ }^{2} D^{e}$ | 3/2 | 0.4316899 |
| 17 | $3 d^{5}$ | ${ }^{4} P^{e}$ | 3/2 | 0.1988758 | 81 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} F^{o}$ | 9/2 | 0.4325285 |
| 18 | $3 d^{5}$ | ${ }^{4} D^{e}$ | 7/2 | 0.2281237 | 82 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} F^{o}$ | 11/2 | 0.4351444 |
| 19 | $3 d^{5}$ | ${ }^{4} D^{e}$ | 1/2 | 0.2281392 | 83 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} P^{o}$ | 3/2 | 0.4410433 |
| 20 | $3 d^{5}$ | ${ }^{4} D^{e}$ | 3/2 | 0.2282067 | 84 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} P^{o}$ | 5/2 | 0.4418831 |
| 21 | $3 d^{5}$ | ${ }^{4} D^{e}$ | 5/2 | 0.2282427 | 85 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} P^{o}$ | 7/2 | 0.4431681 |
| 22 | $3 d^{4}(3 P) 4 s$ | ${ }^{4} P^{e}$ | 1/2 | 0.2729414 | 86 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} P^{o}$ | 1/2 | 0.4442365 |
| 23 | $3 d^{5}$ | ${ }^{2} I^{e}$ | 11/2 | 0.2746857 | 87 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} P^{o}$ | 3/2 | 0.4465745 |
| 24 | $3 d^{5}$ | ${ }^{2} I^{e}$ | 13/2 | 0.2747453 | 88 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} D^{o}$ | 5/2 | 0.4497263 |
| 25 | $3 d^{4}(3 H) 4 s$ | ${ }^{4} H^{e}$ | 7/2 | 0.2748087 | 89 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} D^{\circ}$ | 1/2 | 0.4510110 |
| 26 | $3 d^{4}(3 H) 4 s$ | ${ }^{4} H^{e}$ | 9/2 | 0.2753738 | 90 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} D^{o}$ | 3/2 | 0.4516655 |
| 27 | $3 d^{4}(3 H) 4 s$ | ${ }^{4} H^{e}$ | 11/2 | 0.2761001 | 91 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} D^{o}$ | 7/2 | 0.4524052 |
| 28 | $3 d^{4}(3 P) 4 s$ | ${ }^{4} P^{e}$ | 3/2 | 0.2761815 | 92 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} P^{o}$ | 5/2 | 0.4529571 |
| 29 | $3 d^{4}(3 H) 4 s$ | ${ }^{4} H^{e}$ | 13/2 | 0.2769505 | 93 | $3 d^{4}(5 D) 4 p$ | ${ }^{6} D^{o}$ | 9/2 | 0.4541604 |
| 30 | $3 d^{4}(3 P) 4 s$ | ${ }^{4} P^{e}$ | 5/2 | 0.2812574 | 94 | $3 d^{4}(1 F) 4 s$ | ${ }^{2} F^{e}$ | 7/2 | 0.4617135 |
| 31 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 3/2 | 0.2832484 | 95 | $3 d^{4}(1 F) 4 s$ | ${ }^{2} F^{e}$ | 5/2 | 0.4618992 |
| 32 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 5/2 | 0.2835623 | 96 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} F^{o}$ | 3/2 | 0.4700690 |
| 33 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 7/2 | 0.2840288 | 97 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} F^{o}$ | 5/2 | 0.4708466 |
| 34 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 9/2 | 0.2844914 | 98 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} F^{o}$ | 7/2 | 0.4719346 |
| 35 | $3 d^{5}$ | ${ }^{2} D^{e}$ | 5/2 | 0.2856902 | 99 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} F^{o}$ | 9/2 | 0.4733363 |
| 36 | $3 d^{5}$ | ${ }^{2} D^{e}$ | 3/2 | 0.2873336 | 100 | $3 d^{5}$ | ${ }^{2} G^{e}$ | 7/2 | 0.4765723 |
| 37 | $3 d^{5}$ | ${ }^{2} F^{e}$ | 7/2 | 0.2948464 | 101 | $3 d^{5}$ | ${ }^{2} G^{e}$ | 9/2 | 0.4767838 |
| 38 | $3 d^{5}$ | ${ }^{2} F^{e}$ | 5/2 | 0.2971038 | 102 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} D^{o}$ | 1/2 | 0.4958931 |
| 39 | $3 d^{5}$ | ${ }^{4} F^{e}$ | 7/2 | 0.2992296 | 103 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} D^{o}$ | 3/2 | 0.4966358 |
| 40 | $3 d^{5}$ | ${ }^{4} F^{e}$ | 3/2 | 0.2993032 | 104 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} D^{o}$ | 5/2 | 0.4977849 |
| 41 | $3 d^{5}$ | ${ }^{4} F^{e}$ | 9/2 | 0.2993903 | 105 | $3 d^{4}(5 D) 4 p$ | ${ }^{4} D^{o}$ | 7/2 | 0.4992325 |
| 42 | $3 d^{5}$ | ${ }^{4} F^{e}$ | 5/2 | 0.2993961 | 106 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 9/2 | 0.4999901 |
| 43 | $3 d^{4}(3 G) 4 s$ | ${ }^{4} G^{e}$ | 5/2 | 0.3045269 | 107 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 3/2 | 0.4999993 |
| 44 | $3 d^{4}(3 G) 4 s$ | ${ }^{4} G^{e}$ | 7/2 | 0.3054666 | 108 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 5/2 | 0.5001352 |
| 45 | $3 d^{4}(3 G) 4 s$ | ${ }^{4} G^{e}$ | 9/2 | 0.3063581 | 109 | $3 d^{4}(3 F) 4 s$ | ${ }^{4} F^{e}$ | 7/2 | 0.5001756 |
| 46 | $3 d^{4}(3 G) 4 s$ | ${ }^{4} G^{e}$ | 11/2 | 0.3070435 | 110 | $3 d^{4}(3 P) 4 s$ | ${ }^{4} P^{e}$ | 5/2 | 0.5014070 |
| 47 | $3 d^{4}(3 H) 4 s$ | ${ }^{2} H^{e}$ | 9/2 | 0.3155802 | 111 | $3 d^{4}(3 P) 4 s$ | ${ }^{4} P^{e}$ | 3/2 | 0.5048301 |
| 48 | $3 d^{4}(3 P) 4 s$ | ${ }^{2} \mathrm{P}^{e}$ | 1/2 | 0.3158387 | 112 | $3 d^{4}(3 P) 4 s$ | ${ }^{4} P^{e}$ | 1/2 | 0.5069029 |
| 49 | $3 d^{4}(3 H) 4 s$ | ${ }^{2} H^{e}$ | 11/2 | 0.3172387 | 113 | $3 d^{4}(3 P) 4 s$ | ${ }^{2} P^{e}$ | 3/2 | 0.5388351 |
| 50 | $3 d^{4}(3 P) 4 s$ | ${ }^{2} P^{e}$ | $3 / 2$ | 0.3221863 | 114 | $3 d^{4}(3 P) 4 s$ | ${ }^{2} P^{e}$ | 1/2 | 0.5424471 |
| 51 | $3 d^{4}(3 F) 4 s$ | ${ }^{2} F^{e}$ | 5/2 | 0.3241302 | 115 | $3 d^{4}(3 F) 4 s$ | ${ }^{2} F^{e}$ | 7/2 | 0.5428423 |
| 52 | $3 d^{4}(3 F) 4 s$ | ${ }^{2} F^{e}$ | 7/2 | 0.3244792 | 116 | $3 d^{4}(3 F) 4 s$ | ${ }^{2} F^{e}$ | 5/2 | 0.5429112 |
| 53 | $3 d^{5}$ | ${ }^{2} H^{e}$ | 9/2 | 0.3245051 | 117 | $3 d^{4}(3 G) 4 s$ | ${ }^{2} G^{e}$ | 9/2 | 0.5712633 |
| 54 | $3 d^{5}$ | ${ }^{2} H^{e}$ | 11/2 | 0.3253903 | 118 | $3 d^{4}(3 G) 4 s$ | ${ }^{2} G^{e}$ | 7/2 | 0.5713792 |
| 55 | $3 d^{5}$ | ${ }^{2} G^{e}$ | 7/2 | 0.3289816 | 119 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} H^{o}$ | 7/2 | 0.5795737 |
| 56 | $3 d^{5}$ | ${ }^{2} G^{e}$ | 9/2 | 0.3305395 | 120 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} H^{o}$ | 9/2 | 0.5805341 |
| 57 | $3 d^{4}(3 D) 4 s$ | ${ }^{4} D^{e}$ | 7/2 | 0.3487379 | 121 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} D^{o}$ | 1/2 | 0.5814051 |
| 58 | $3 d^{4}(3 D) 4 s$ | ${ }^{4} D^{e}$ | 5/2 | 0.3491504 | 122 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} H^{o}$ | 11/2 | 0.5818321 |
| 59 | $3 d^{4}(3 D) 4 s$ | ${ }^{4} D^{e}$ | 3/2 | 0.3495839 | 123 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} H^{o}$ | 13/2 | 0.5834887 |
| 60 | $3 d^{4}(3 D) 4 s$ | ${ }^{4} D^{e}$ | 1/2 | 0.3498919 | 124 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} D^{o}$ | 3/2 | 0.5837730 |
| 61 | $3 d^{4}(3 G) 4 s$ | ${ }^{2} G^{e}$ | 7/2 | 0.3509189 | 125 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} D^{o}$ | 5/2 | 0.5873003 |
| 62 | $3 d^{4}(3 G) 4 s$ | ${ }^{2} G^{e}$ | 9/2 | 0.3514117 | 126 | $3 d^{5} *$ | ${ }^{2} P^{e}$ | 3/2 | 0.5914730 |
| 63 | $3 d^{4}(a 1 G) 4 s$ | ${ }^{2} G^{e}$ | 7/2 | 0.3616247 | 127 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} D^{o}$ | 7/2 | 0.5916353 |
| 64 | $3 d^{5}$ | ${ }^{2} F^{e}$ | 5/2 | 0.3621563 | 128 | $3 d^{5} *$ | ${ }^{2} P^{e}$ | 1/2 | 0.5924667 |

Table 1
(Continued)

| Index | Config. | LS | $J$ | Energy | Index | Config. | LS | $J$ | Energy |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 129 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{2} S^{o}$ | 1/2 | 0.5925913 | 193 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} D^{o}$ | 7/2 | 0.6696501 |
| 130 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} G^{o}$ | 5/2 | 0.5937498 | 194 | $3 d^{4}(a 1 G) 4 p$ | ${ }^{2} F^{o}$ | 7/2 | 0.6753795 |
| 131 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} I^{o}$ | 9/2 | 0.5943056 | 195 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} F^{o}$ | 3/2 | 0.6768277 |
| 132 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} G^{o}$ | 7/2 | 0.5946640 | 196 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} F^{o}$ | 5/2 | 0.6772425 |
| 133 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} G^{o}$ | 9/2 | 0.5958219 | 197 | $3 d^{4}(1 I) 4 p$ | ${ }^{2} I^{O}$ | 11/2 | 0.6781808 |
| 134 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} I^{o}$ | 11/2 | 0.5961464 | 198 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} F^{o}$ | 7/2 | 0.6781979 |
| 135 | $3 d^{4}(3 H) 4 p$ | ${ }^{2} G^{o}$ | 7/2 | 0.5972707 | 199 | $3 d^{4}(1 I) 4 p$ | ${ }^{2} K^{o}$ | 13/2 | 0.6782026 |
| 136 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} I^{o}$ | 13/2 | 0.5979547 | 200 | $3 d^{4}(a 1 G) 4 p$ | ${ }^{2} F^{o}$ | 5/2 | 0.6783116 |
| 137 | $3 d^{4}(3 H) 4 p$ | ${ }^{2} G^{o}$ | 9/2 | 0.5985205 | 201 | $3 d^{4}(a 1 G) 4 p$ | ${ }^{2} H^{o}$ | $9 / 2$ | 0.6784908 |
| 138 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} G^{o}$ | 11/2 | 0.5987886 | 202 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} P^{o}$ | 5/2 | 0.6787473 |
| 139 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} I^{o}$ | 15/2 | 0.5997277 | 203 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} F^{o}$ | $9 / 2$ | 0.6789314 |
| 140 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} P^{o}$ | 1/2 | 0.6037729 | 204 | $3 d^{4}(a 1 G) 4 p$ | ${ }^{2} H^{o}$ | 11/2 | 0.6807841 |
| 141 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} P^{o}$ | 3/2 | 0.6046692 | 205 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} P^{o}$ | 3/2 | 0.6808763 |
| 142 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{2} P^{o}$ | 3/2 | 0.6073534 | 206 | $3 d^{4}(1 I) 4 p$ | ${ }^{2} I^{o}$ | 13/2 | 0.6811104 |
| 143 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} P^{o}$ | 5/2 | 0.6080590 | 207 | $3 d^{4}(3 D) 4 p$ | ${ }^{2} P^{o}$ | 1/2 | 0.6821179 |
| 144 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{2} P^{o}$ | 1/2 | 0.6093814 | 208 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} P^{o}$ | $1 / 2$ | 0.6827248 |
| 145 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} F^{o}$ | 5/2 | 0.6106587 | 209 | $3 d^{4}(1 I) 4 p$ | ${ }^{2} K^{o}$ | 15/2 | 0.6830743 |
| 146 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} F^{o}$ | 3/2 | 0.6111904 | 210 | $3 d^{4}(3 D) 4 p$ | ${ }^{2} P^{o}$ | 3/2 | 0.6833111 |
| 147 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} G^{o}$ | 7/2 | 0.6135905 | 211 | $3 d^{4}(a 1 G) 4 p$ | ${ }^{2} G^{o}$ | 7/2 | 0.6899794 |
| 148 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} G^{o}$ | 5/2 | 0.6136835 | 212 | $3 d^{4}(a 1 G) 4 p$ | ${ }^{2} G^{o}$ | $9 / 2$ | 0.6908314 |
| 149 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} G^{o}$ | 9/2 | 0.6137679 | 213 | $3 d^{4}(3 D) 4 p$ | ${ }^{2} F^{o}$ | 7/2 | 0.7005727 |
| 150 | $3 d^{4}(3 H) 4 p$ | ${ }^{4} G^{o}$ | 11/2 | 0.6139123 | 214 | $3 d^{4}(3 D) 4 p$ | ${ }^{2} F^{o}$ | 5/2 | 0.7015630 |
| 151 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{2} D^{o}$ | 3/2 | 0.6140064 | 215 | $3 d^{4}(1 I) 4 p$ | ${ }^{2} H^{o}$ | 11/2 | 0.7023943 |
| 152 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{2} D^{o}$ | 5/2 | 0.6140765 | 216 | $3 d^{4}(1 I) 4 p$ | ${ }^{2} \mathrm{H}^{\circ}$ | $9 / 2$ | 0.7041378 |
| 153 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} F^{o}$ | 7/2 | 0.6141344 | 217 | $3 d^{4}(a 1 S) 4 p$ | ${ }^{2} P^{o}$ | 3/2 | 0.7081750 |
| 154 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} F^{o}$ | 9/2 | 0.6146361 | 218 | $3 d^{4}(a 1 S) 4 p$ | ${ }^{2} P^{o}$ | $1 / 2$ | 0.7087586 |
| 155 | $3 d^{4}(3 H) 4 p$ | ${ }^{2} I^{o}$ | 11/2 | 0.6151606 | 219 | $3 d^{4}(3 D) 4 p$ | ${ }^{2} D^{o}$ | 5/2 | 0.7101977 |
| 156 | $3 d^{4}(3 H) 4 p$ | ${ }^{2} I^{o}$ | 13/2 | 0.6159149 | 220 | $3 d^{4}(3 D) 4 p$ | ${ }^{2} D^{o}$ | $3 / 2$ | 0.7117861 |
| 157 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} D^{o}$ | 1/2 | 0.6183818 | 221 | $3 d^{4}(a 1 D) 4 p$ | ${ }^{2} D^{o}$ | 3/2 | 0.7316380 |
| 158 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} D^{o}$ | 5/2 | 0.6184566 | 222 | $3 d^{4}(a 1 D) 4 p$ | ${ }^{2} D^{o}$ | 5/2 | 0.7328422 |
| 159 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} D^{o}$ | 3/2 | 0.6184787 | 223 | $3 d^{4}(a 1 D) 4 p$ | ${ }^{2} F^{o}$ | 5/2 | 0.7402483 |
| 160 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{4} D^{o}$ | 7/2 | 0.6185259 | 224 | $3 d^{4}(a 1 D) 4 p$ | ${ }^{2} F^{o}$ | 7/2 | 0.7420651 |
| 161 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{4} S^{o}$ | 3/2 | 0.6224463 | 225 | $3 d^{4}(a 1 D) 4 p$ | ${ }^{2} P^{o}$ | $1 / 2$ | 0.7550193 |
| 162 | $3 d^{4}(3 H) 4 p$ | ${ }^{2} H^{o}$ | 9/2 | 0.6240071 | 226 | $3 d^{4}(a 1 D) 4 p$ | ${ }^{2} P^{o}$ | 3/2 | 0.7556211 |
| 163 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{2} F^{o}$ | 5/2 | 0.6249769 | 227 | $3 d^{4}(1 F) 4 p$ | ${ }^{2} F^{o}$ | 5/2 | 0.7709754 |
| 164 | $3 d^{4}(3 H) 4 p$ | ${ }^{2} \mathrm{H}^{\circ}$ | 11/2 | 0.6263846 | 228 | $3 d^{4}(1 F) 4 p$ | ${ }^{2} F^{o}$ | 7/2 | 0.7716342 |
| 165 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{2} F^{o}$ | 7/2 | 0.6265857 | 229 | $3 d^{4}(1 F) 4 p$ | ${ }^{2} G^{o}$ | 7/2 | 0.7797995 |
| 166 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} H^{o}$ | 7/2 | 0.6273458 | 230 | $3 d^{4}(1 F) 4 p$ | ${ }^{2} G^{o}$ | $9 / 2$ | 0.7831328 |
| 167 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} H^{o}$ | 9/2 | 0.6287045 | 231 | $3 d^{4}(1 F) 4 p$ | ${ }^{2} D^{o}$ | 5/2 | 0.7883120 |
| 168 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} H^{o}$ | 11/2 | 0.6303265 | 232 | $3 d^{4} 4 s *$ | ${ }^{2} D^{e}$ | $3 / 2$ | 0.7889768 |
| 169 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{2} D^{o}$ | 3/2 | 0.6319466 | 233 | $3 d^{4} 4 s *$ | ${ }^{2} D^{e}$ | 5/2 | 0.7889804 |
| 170 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} H^{o}$ | 13/2 | 0.6323118 | 234 | $3 d^{4}(1 F) 4 p$ | ${ }^{2} D^{o}$ | 3/2 | 0.7920637 |
| 171 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} F^{o}$ | 5/2 | 0.6331292 | 235 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} P^{o}$ | 3/2 | 0.8148759 |
| 172 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} F^{o}$ | 9/2 | 0.6333144 | 236 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} P^{o}$ | 5/2 | 0.8151570 |
| 173 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} F^{o}$ | 7/2 | 0.6333855 | 237 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} P^{o}$ | 1/2 | 0.8156557 |
| 174 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} F^{o}$ | 3/2 | 0.6345934 | 238 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} D^{o}$ | 7/2 | 0.8221307 |
| 175 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{2} G^{o}$ | 7/2 | 0.6370075 | 239 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} D^{o}$ | 5/2 | 0.8224926 |
| 176 | $3 d^{4}(a 3 P) 4 p$ | ${ }^{2} D^{o}$ | 5/2 | 0.6374681 | 240 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} F^{o}$ | 3/2 | 0.8225284 |
| 177 | $3 d^{4}(a 3 F) 4 p$ | ${ }^{2} G^{o}$ | 9/2 | 0.6388676 | 241 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} F^{o}$ | 5/2 | 0.8241655 |
| 178 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} G^{o}$ | 5/2 | 0.6407742 | 242 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} D^{o}$ | 3/2 | 0.8242461 |
| 179 | $3 d^{4}(3 G) 4 p$ | ${ }^{2} H^{o}$ | 9/2 | 0.6414787 | 243 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} D^{o}$ | 1/2 | 0.8244726 |
| 180 | $3 d^{4}(3 G) 4 p$ | ${ }^{2} H^{\circ}$ | 11/2 | 0.6415212 | 244 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} F^{o}$ | 7/2 | 0.8246036 |
| 181 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} G^{o}$ | 7/2 | 0.6417780 | 245 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} F^{o}$ | 9/2 | 0.8255033 |
| 182 | $3 d^{4}(3 G) 4 p$ | ${ }^{2} F^{o}$ | 5/2 | 0.6432125 | 246 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{2} F^{o}$ | 5/2 | 0.8265807 |
| 183 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} G^{o}$ | 9/2 | 0.6440753 | 247 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{2} F^{o}$ | 7/2 | 0.8277104 |
| 184 | $3 d^{4}(3 G) 4 p$ | ${ }^{2} F^{o}$ | 7/2 | 0.6456525 | 248 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} G^{o}$ | 5/2 | 0.8299697 |
| 185 | $3 d^{4}(3 G) 4 p$ | ${ }^{4} G^{o}$ | 11/2 | 0.6459038 | 249 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} G^{o}$ | 7/2 | 0.8303716 |
| 186 | $3 d^{5} *$ | ${ }^{2} D^{e}$ | 5/2 | 0.6527753 | 250 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} G^{o}$ | $9 / 2$ | 0.8309793 |
| 187 | $3 d^{5} *$ | ${ }^{2} D^{e}$ | 3/2 | 0.6529060 | 251 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} G^{o}$ | 11/2 | 0.8319147 |
| 188 | $3 d^{4}(3 G) 4 p$ | ${ }^{2} G^{o}$ | 7/2 | 0.6620215 | 252 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{2} D^{o}$ | 5/2 | 0.8331349 |
| 189 | $3 d^{4}(3 G) 4 p$ | ${ }^{2} G^{o}$ | 9/2 | 0.6626430 | 253 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{2} D^{o}$ | 3/2 | 0.8343227 |
| 190 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} D^{o}$ | 1/2 | 0.6689333 | 254 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{4} S^{o}$ | 3/2 | 0.8439445 |
| 191 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} D^{o}$ | 3/2 | 0.6689785 | 255 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} D^{o}$ | 7/2 | 0.8523228 |
| 192 | $3 d^{4}(3 D) 4 p$ | ${ }^{4} D^{o}$ | 5/2 | 0.6691991 | 256 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{2} G^{o}$ | $9 / 2$ | 0.8533235 |

Table 1
(Continued)

| Index | Config. | LS | $J$ | Energy | Index | Config. | LS | $J$ | Energy |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 257 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} D^{o}$ | $5 / 2$ | 0.8535923 | 269 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{2} D^{o}$ | $3 / 2$ | 0.8959114 |
| 258 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} D^{o}$ | $3 / 2$ | 0.8544954 | 270 | $3 d^{4}(b 1 G) 4 p$ | ${ }^{2} F^{o}$ | $7 / 2$ | 0.9027851 |
| 259 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{4} D^{o}$ | $1 / 2$ | 0.8547714 | 271 | $3 d^{4}(b 1 G) 4 p$ | ${ }^{2} F^{o}$ | $5 / 2$ | 0.9043771 |
| 260 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{2} G^{o}$ | $7 / 2$ | 0.8547811 | 272 | $3 d^{4} 4 s *$ | ${ }^{2} P^{e}$ | $1 / 2$ | 0.9805123 |
| 261 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{2} P^{o}$ | $3 / 2$ | 0.8600823 | 273 | $3 d^{4} 4 p *$ | ${ }^{2} P^{o}$ | $3 / 2$ | 1.0399800 |
| 262 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{2} P^{o}$ | $1 / 2$ | 0.8622832 | 274 | $3 d^{4} 4 p *$ | ${ }^{2} P^{o}$ | $1 / 2$ | 1.0422870 |
| 263 | $3 d^{4}(b 3 P) 4 p$ | ${ }^{2} S^{o}$ | $1 / 2$ | 0.8770512 | 275 | $3 d^{4} 4 p *$ | ${ }^{2} D^{o}$ | $3 / 2$ | 1.0515680 |
| 264 | $3 d^{4}(b 1 G) 4 p$ | ${ }^{2} H^{o}$ | $9 / 2$ | 0.8883032 | 276 | $3 d^{4} 4 p *$ | ${ }^{2} F^{o}$ | $5 / 2$ | 1.0560210 |
| 265 | $3 d^{4}(b 1 G) 4 p$ | ${ }^{2} G^{o}$ | $7 / 2$ | 0.8905647 | 277 | $3 d^{4} 4 p *$ | ${ }^{2} F^{o}$ | $5 / 2$ | 1.0667810 |
| 266 | $3 d^{4}(b 1 G) 4 p$ | ${ }^{2} H^{o}$ | $11 / 2$ | 0.8921244 | 278 | $3 d^{4} 4 p *$ | ${ }^{2} G^{o}$ | $7 / 2$ | 1.0701220 |
| 267 | $3 d^{4}(b 1 G) 4 p$ | ${ }^{2} G^{o}$ | $9 / 2$ | 0.8921702 | 279 | $3 d^{4} 4 p *$ | ${ }^{2} S^{o}$ | $1 / 2$ | 1.1736580 |
| 268 | $3 d^{4}(b 3 F) 4 p$ | ${ }^{2} D^{o}$ | $5 / 2$ | 0.8949306 | 280 | $3 d^{4} 4 p *$ | ${ }^{2} D^{o}$ | $3 / 2$ | 1.1736820 |

Note. * represents unobserved levels.
were included in the wavefunction representation of the Cr II ion. These states were represented by configuration-interaction wavefunction expansions in terms of nine orthogonal basis orbitals, eight spectroscopic ( $1 s, 2 s, 2 p, 3 s, 3 p, 3 d, 4 s$, and $4 p$ ) and one pseudo-type corrector orbital $(\overline{4 d})$ included to account for additional correlation effects. The parameters utilized for the Hartree-Fock orbitals were taken from the tables of Clementi \& Roetti (1974) while those for the remaining non-Hartree-Fock orbitals were obtained variationally using the CIV3 code of Hibbert (1975). All parameter values which describe the orbital basis can be found in the tabulations of Wasson et al. (2010) and will not be reproduced here. In Table 1, we present the target state energies in Rydbergs relative to the $3 d^{5}{ }^{6} S_{5 / 2}^{e}$ ground state for all 280 fine-structure levels considered. These energies have been shifted to their observed experimental values listed in the NIST compilation where available, to ensure that thresholds lie in their exact positions. In Table 1, each target level is assigned an index value ranging from 1 to 280 which will be used again in subsequent tables when denoting a particular transition.

The RMATRX II $R$-matrix suite of codes (Burke et al. 1994) were used in the present calculation to compute the diagonalized Hamiltonian matrices in the internal region using LS coupling. Relativistic effects were incorporated by transforming the $R$-matrix in LS coupling at energy $E$ into an $R$-matrix in pair coupling (private communication with V. M. Burke in association with Sunderland et al. 2002). In this method the energy independent surface amplitudes are transformed at the $R$-matrix boundary and the term splitting in the target is taken into account by employing the term coupling coefficients. One of the advantages of using this method, over other transformation procedures, is that it allows us to include fine-structure channels in the external region as the transformation occurs on the $R$-matrix boundary as opposed to the asymptotic boundary. Finally, the PSTGF external region code of Ballance \& Griffin (2004) was used to compute the collision strengths over a very fine mesh of incident electron energies to ensure proper delineation of the complex resonance structures which dominate the low-energy collision cross sections.

A full exchange calculation was carried out for this $280 j j$ level approximation for all singlet, triplet, and quintet spin symmetries with total angular momentum $2 J \leqslant 14$. This was further augmented with a non-exchange calculation to account for higher partial wave contributions. Due to the long range nature of the Coulomb potential, a further contribution to the dipole allowed transitions come from even higher partial waves.


Figure 1. Collision strength against incident electron energy, in Rydbergs, for transitions from the ground state, $3 d^{5}{ }^{6} S_{5 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{6} P_{J}^{o}$ for $J=3 / 2,5 / 2$, and $7 / 2(1-83,1-84$, and $1-85)$, respectively. The solid black line corresponds to the present work, the dot-dashed green line corresponds to Model 2J14, and the dashed red line is the work of Bautista et al. (2009).
(A color version of this figure is available in the online journal.)

We incorporate this "top-up" contribution using the Burgess sum rule for the dipole transitions and a geometric series for the non-dipole transitions, with care being taken to ensure smooth convergence toward the high-energy limit. These important high partial wave contributions were found to be negligible for the low-lying forbidden lines of Cr II considered by Wasson et al. (2010), but as we shall show in the results section are significant for the optically allowed transitions of interest in the present paper.

## 3. RESULTS AND DISCUSSION

We present in Figures 1-9 the collision strengths, $\Omega_{i j}$, and Maxwellian averaged effective collision strengths, $\Upsilon_{i j}$, for a selection of dipole allowed transitions in Cr II to investigate convergence and the effects of high partial wave contributions particularly at high incident electron energies. We note that the transitions are labeled according to the index numbers $i$ and $j$ assigned in Table 1. The effective collision strengths are of importance in astrophysical and plasma applications and are obtained by averaging over a Maxwellian distribution of electron


Figure 2. Effective collision strength against electron temperature in Kelvin (log scale), for transitions from the ground state, $3 d^{5}{ }^{6} S_{5 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{6} P_{J}^{o}$ for $J=3 / 2,5 / 2$, and $7 / 2(1-83,1-84$, and $1-85$ ), respectively. The solid black line corresponds to the present work, the dot-dashed green line corresponds to Model 2J14, and the dashed red line is the work of Bautista et al. (2009).
(A color version of this figure is available in the online journal.)


Figure 3. Reduced collision strength against reduced electron temperature, for transitions from the ground state, $3 d^{5}{ }^{6} S_{5 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{6} P_{J}^{o}$ for $J=3 / 2,5 / 2$, and $7 / 2(1-83,1-84$, and 1-85), respectively.
(A color version of this figure is available in the online journal.)
velocities so that

$$
\begin{equation*}
\Upsilon_{i j}=\int_{0}^{\infty} \Omega_{i j}\left(E_{j}\right) \exp \left(-E_{j} / k T\right) d\left(E_{j} / k T\right) \tag{1}
\end{equation*}
$$

Collision strength and effective collision strength comparisons are made with the theoretical work of Bautista et al. (2009) who included all partial waves with angular momentum $L=0-10$, with higher contributions accounted for via a Coulomb Bethe "top-up" procedure (Burgess 1974). In this work the collision strengths for the fine-structure levels were obtained by algebraic recoupling of the LS reaction matrices (Hummer et al. 1993). The present work includes all partial waves up to and including $2 J=26$ with higher partial wave contributions included via a similar "top-up" procedure. In an attempt to thoroughly investigate the effect of higher partial wave contributions, top-up and convergence of the collision strengths and subsequently the


Figure 4. Collision strength against incident electron energy, in Rydbergs, for transitions from the excited state, $3 d^{4} 4 s^{6} D_{7 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{6} D_{J}^{o}$ for $J=5 / 2,7 / 2$, and $9 / 2$ (5-88, 5-91, and $5-93$ ), respectively. The solid black line corresponds to the present work, the dot-dashed green line corresponds to Model 2J14 and the dashed red line is the work of Bautista et al. (2009).
(A color version of this figure is available in the online journal.)


Figure 5. Effective collision strength against electron temperature in Kelvin (log scale), for transitions from the excited state, $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{6} D_{J}^{o}$ for $J=5 / 2,7 / 2$, and $9 / 2$ (5-88, 5-91, and 5-93), respectively. The solid black line corresponds to the present work, the dot-dashed green line corresponds to Model 2J14, and the dashed red line is the work of Bautista et al. (2009).
(A color version of this figure is available in the online journal.)
effective collision strengths, we have also included in our figures the total partial wave contributions up to and including $2 J=14$. This model is referred to as "Model 2J14" and corresponds to the data which would have been produced by the Wasson et al. (2010) investigation for these allowed lines. We remember from this work that convergence of the forbidden lines was the priority and the allowed transitions were not considered. A comparison of data between Model 2J14 and the present work will allow us to thoroughly investigate the contributions from partial waves corresponding to $2 J$ values higher than 14 .

We consider first transitions from the ground $3 d^{5}{ }^{6} S_{5 / 2}^{e}$ state of Cr II to the odd parity $3 d^{4} 4 p{ }^{6} P_{J}^{o}$ state with split levels $J=3 / 2,5 / 2,7 / 2$ (transitions $1-83,1-84$, and $1-85$ ). In Figure 1, we present the collision strength as a function of incident electron energy in Rydbergs for all three dipole allowed


Figure 6. Reduced collision strength against reduced electron temperature, for transitions from the excited state, $3 d^{4} 4 s^{6} D_{7 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p{ }^{6} D_{J}^{o}$ for $J=5 / 2,7 / 2$, and $9 / 2(5-88,5-91$, and 5-93), respectively.
(A color version of this figure is available in the online journal.)


Figure 7. Collision strength against incident electron energy, in Rydbergs, for transitions from the excited state, $3 d^{4} 4 s^{4} D_{3 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{4} P_{J}^{o}$ for $J=1 / 2,3 / 2$, and $5 / 2(8-86,8-87$, and $8-92$ ), respectively. The solid black line corresponds to the present work, the dot-dashed green line corresponds to Model 2J14, and the dashed red line is the work of Bautista et al. (2009).
(A color version of this figure is available in the online journal.)
transitions. The present work and the theoretical cross sections of Bautista et al. (2009) are in excellent agreement up to the highest incident energy of 1 Rydberg. Beyond this energy only the present cross sections are evaluated up to approximately 5 Rydbergs. Differences of only a few percent are recorded for all three transitions between these two works. The collision strengths of Model 2 J 14 lie, as expected, significantly below the other two, emphasizing the importance of high partial wave contributions and "top-up" for these dipole allowed transitions.

In Figure 2, we present the Maxwellian averaged collision strengths as a function of electron temperature in Kelvin on a logarithmic scale, for the same three transitions. For the lowest temperatures considered ( $T=2000 \rightarrow 10,000 \mathrm{~K}$ ) excellent agreement is evident between the present results, Bautista et al. (2009) and Model 2J14. Beyond this temperature value deviations occur and at the highest temperature considered the present


Figure 8. Effective collision strength against electron temperature in Kelvin (log scale), for transitions from the excited state, $3 d^{4} 4 s{ }^{4} D_{3 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{4} P_{J}^{o}$ for $J=1 / 2,3 / 2$, and 5/2 ( $8-86,8-87$, and $8-92$ ), respectively. The solid black line corresponds to the present work, the dot-dashed green line corresponds to Model 2J14, and the dashed red line is the work of Bautista et al. (2009).
(A color version of this figure is available in the online journal.)


Figure 9. Reduced collision strength against reduced electron temperature, for transitions from the excited state, $3 d^{4} 4 s^{4} D_{3 / 2}^{e}$ to the three fine-structure levels of the excited state $3 d^{4} 4 p^{4} P_{J}^{o}$ for $J=1 / 2,3 / 2$, and $5 / 2(8-86,8-87$, and 8-92), respectively.
(A color version of this figure is available in the online journal.)
work indicates effective collision strengths significantly higher than the other predictions. We note the highest temperature considered by Bautista et al. (2009) is $T=30,000 \mathrm{~K}$.

As a further check on the accuracy of the atomic data presented in Figures 1 and 2, we can ascertain that our results conform to the expected infinite-energy limits. We utilize the method of Burgess \& Tully (1992) where a reduced collision strength, $\Omega_{r}$ is plotted against a reduced energy, $E_{r}$, mapping the complete range of incident electron energies onto the interval $(0,1)$. The reduced energy $E_{r}$ is calculated as

$$
\begin{equation*}
E_{r}=1-\frac{\ln C}{\ln \left(\frac{E_{j}}{\Delta E}+C\right)} \tag{2}
\end{equation*}
$$

where $C$ is an adjustable parameter permitting flexibility, $E_{j}$ is the incident electron energy following excitation, and $\Delta E$ is

Table 2
Oscillator Strengths $g f_{i j}$ Values for Transitions in Cr II Considered in the Figures

| $i$ | ${ }^{j}$ | $g f_{i j}$ |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Present Work | Nilsson et al. (2006) | Spreger et al. <br> (1994) | Gonzalez et al. (1994) | Bergeson \& Lawler (1993) | Musielok \& Wujec (1979) | Aashamar \& Luke (1990) | Kurucz <br> (1988) |
| 1 | 83 | 0.48136 | 0.30061 | $\ldots$ | 0.30479 | 0.30903 | $\ldots$ | 0.48000 | 0.50466 |
|  | 84 | 0.71946 | 0.48753 | $\ldots$ | 0.45082 | 0.46774 | ... | 0.72900 | 0.76913 |
|  | 85 | 0.95880 | 0.65163 | $\ldots$ | 0.60814 | 0.63096 | ... | 0.99100 | 1.05925 |
| 5 | 88 | 0.64282 | 0.28708 | ... | ... | ... | ... | 0.37900 | 0.08375 |
|  | 91 | 1.07536 | 1.02565 | ... | $\ldots$ | ... | $\ldots$ | 1.02100 | 1.09144 |
|  | 93 | 0.56977 | 0.56105 | $\ldots$ | $\ldots$ | $\ldots$ | ... | 0.52000 | 0.62373 |
| 8 | 86 | 0.16673 | 0.18664 | $\ldots$ | ... | $\ldots$ | 0.45920 | ... | 0.23768 |
|  | 87 | 0.21226 | 0.21677 | ... | $\ldots$ | ... | ... | ... | 0.21677 |
|  | 92 | 0.03481 | 0.02958 | 0.06166 | $\ldots$ | $\ldots$ | ... | $\ldots$ | 0.02037 |

Notes. Present work values are taken from the length formula calculations in Table 3. Comparisons are made with the experimental works of Nilsson et al. (2006), Spreger et al. (1994), Gonzalez et al. (1994), Bergeson \& Lawler (1993), and Musielok \& Wujec (1979) and, finally, the theoretical works of Aashamar \& Luke (1990) and Kurucz (1988).
the transition energy between the excited states. The reduced collision strength as a function of the reduced energy is represented by

$$
\begin{equation*}
\Omega_{r}\left(E_{r}\right)=\frac{\Omega\left(E_{j}\right)}{\ln \left(\frac{E_{j}}{\Delta E}+e\right)} \tag{3}
\end{equation*}
$$

where $e$ is Euler's constant.
We note that at the infinite-energy limit ( $E_{r}=1$ ) the reduced collision strength behaves asymptotically as

$$
\begin{equation*}
\Omega_{r}\left(E_{r}\right)=\frac{4 g_{i} f_{i j}}{\Delta E} \tag{4}
\end{equation*}
$$

where $g_{i}$ is the statistical weight of the initial level $\left(2 J_{i}+1\right)$ and $f_{i j}$ denotes the absorption oscillator strength. The infinite-energy limit will clearly vary linearly with the choice of oscillator strength. Table 2 compares the oscillator strengths computed in the present work for the nine transitions considered in this paper with the experimental works of Nilsson et al. (2006), Spreger et al. (1994), Gonzalez et al. (1994), Bergeson \& Lawler (1993), and Musielok \& Wujec (1979) and, finally, the theoretical works of Aashamar \& Luke (1990) and Kurucz (1988).

In Figure 3, we present the reduced collision strength $\Omega_{r}$ as a function of reduced incident electron energy $E_{r}$ for the three transitions $3 d^{5}{ }^{6} S_{5 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{6} P_{3 / 2,5 / 2,7 / 2}^{o}$ (transitions 1-83, $1-84$, and $1-85$ ) considered in Figures 1 and 2. Comparisons are made with the reduced plots formed using the variety of available theoretical and experimental oscillator strengths in Table 2. From Figure 3, we see that the infinite-energy limits predicted by the present work are in very good agreement with the predictions using the $f$-values of Kurucz (1988) and Aashamar \& Luke (1990) but slightly higher than those evaluated using the Nilsson et al. (2006) and Gonzalez et al. (1994) data, for all three transitions.

In Figure 4, we present the collision strength for dipole allowed transitions from the metastable $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e}$ (index 5) to the $3 d^{4} 4 p^{6} D_{J}^{o}$ for $J=5 / 2,7 / 2$, and $9 / 2$ (indices 88,91 , and 93). Again we see the importance of the high partial wave contributions which are absent in Model 2J14. Agreement is not unsatisfactory with the collision strengths of Bautista et al. (2009) for the $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{6} D_{7 / 2,9 / 2}^{o}$ transitions, but his predictions for the $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{6} D_{5 / 2}^{o}$

Table 3
Theoretical Oscillator Strengths, $f$, and Transition Probabilities, $A$, between All Allowed Fine-structure Transitions of Cr II

| $i$ | $j$ | $J(i)$ | $J(j)$ | $\Delta E$ | $f_{\text {length }}$ | $f_{\text {velocity }}$ | $A_{\text {length }}$ | $A_{\text {velocity }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 76 | $5 / 2$ | $3 / 2$ | 0.22120 | $4.8747^{-08}$ | $5.7941^{-08}$ | $1.1495^{+02}$ | $1.3664^{+02}$ |
| 1 | 77 | $5 / 2$ | $5 / 2$ | 0.22223 | $1.7075^{-07}$ | $2.0714^{-07}$ | $2.7095^{+02}$ | $3.2869^{+02}$ |
| 1 | 78 | $5 / 2$ | $7 / 2$ | 0.22365 | $2.0445^{-07}$ | $2.5068^{-07}$ | $2.4643^{+02}$ | $3.0216^{+02}$ |
| 1 | 83 | $5 / 2$ | $3 / 2$ | 0.22762 | $8.0226^{-02}$ | $7.8855^{-02}$ | $2.0032^{+08}$ | $1.9690^{+08}$ |
| 1 | 84 | $5 / 2$ | $5 / 2$ | 0.22870 | $1.1991^{-01}$ | $1.1862^{-01}$ | $2.0150^{+08}$ | $1.9935^{+08}$ |
| 1 | 85 | $5 / 2$ | $7 / 2$ | 0.23021 | $1.5980^{-01}$ | $1.5934^{-01}$ | $2.0408^{+08}$ | $2.0350^{+08}$ |
| 1 | 87 | $5 / 2$ | $3 / 2$ | 0.23884 | $1.7733^{-04}$ | $1.7102^{-04}$ | $4.8751^{+05}$ | $4.7016^{+05}$ |
| 1 | 88 | $5 / 2$ | $5 / 2$ | 0.23720 | $5.7059^{-05}$ | $3.2965^{-05}$ | $1.0315^{+05}$ | $5.9591^{+04}$ |
| 1 | 90 | $5 / 2$ | $3 / 2$ | 0.23509 | $1.7151^{-04}$ | $1.3945^{-04}$ | $4.5683^{+05}$ | $3.7144^{+05}$ |
| 1 | 91 | $5 / 2$ | $7 / 2$ | 0.23903 | $1.5607^{-05}$ | $3.0811^{-05}$ | $2.1489^{+04}$ | $4.2423^{+04}$ |

(This table is available in its entirety in a machine-readable form in the online journal. A portion is shown here for guidance regarding its form and content.)
lie considerably higher than the present work at all incident electron energies. The Maxwellian averaged effective collision strengths presented in Figure 5 for these three transitions corroborate these findings. Agreement is excellent across the temperature range $T=2000-30,000 \mathrm{~K}$ for the $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e} \rightarrow$ $3 d^{4} 4 p{ }^{6} D_{7 / 2,9 / 2}^{o}$ transitions, but the Bautista et al. (2009) data lie significantly higher than the present work for the $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{6} D_{5 / 2}^{o}$ transition. It should also be noted that the present work includes a much more extensive temperature range, $T=2000-100,000 \mathrm{~K}$, than the previous evaluation.

The oscillator strengths for these three transitions are listed in Table 2 where we see excellent agreement between the theories of Kurucz (1988), Aashamar \& Luke (1990), the present work, and the experimental values of Nilsson et al. (2006) for the $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{6} D_{7 / 2,9 / 2}^{o}$ transitions. Differences of less than $2 \%$ are found between the present theory and the experimental data of Nilsson for these two transitions. For the $3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{6} D_{5 / 2}^{o}$ transition, however, experiment and theory do not agree with discrepancies of $50 \%$ or more evident. We clearly see the effect of these differing oscillator strengths in the reduced energy plots of Figure 6 for these transitions. Excellent agreement is found as expected for the lower two transitions whereas the different $f$-values predict different high-energy limits for the

Table 4
Maxwellian Averaged Effective Collision Strengths against Temperature (in Kelvin) for Allowed Fine-structure Transitions of Cr II

| $i$ | ${ }^{j}$ | Temperature (K) |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 2,000 | 2,300 | 2,500 | 5,000 | 7,500 | 10,000 | 13,000 | 15,000 | 18,000 |
|  |  | 20,000 | 30,000 | 40,000 | 50,000 | 60,000 | 70,000 | 80,000 | 90,000 | 100,000 |
| Effective Collision Strength |  |  |  |  |  |  |  |  |  |  |
| 1 | 76 | $3.46{ }^{-01}$ | $3.30^{-01}$ | $3.21^{-01}$ | $2.51{ }^{-01}$ | $2.21^{-01}$ | $2.06{ }^{-01}$ | $1.96{ }^{-01}$ | $1.92{ }^{-01}$ | $1.89^{-01}$ |
|  |  | $1.88{ }^{-01}$ | $1.85{ }^{-01}$ | $1.85{ }^{-01}$ | $1.84{ }^{-01}$ | $1.83{ }^{-01}$ | $1.82{ }^{-01}$ | $1.80^{-01}$ | $1.79{ }^{-01}$ | $1.78{ }^{-01}$ |
| 1 | 77 | $7.82^{-01}$ | $7.26{ }^{-01}$ | $6.95{ }^{-01}$ | $4.89{ }^{-01}$ | $4.11^{-01}$ | $3.71^{-01}$ | $3.44{ }^{-01}$ | $3.32^{-01}$ | $3.19^{-01}$ |
|  |  | $3.13{ }^{-01}$ | $2.97{ }^{-01}$ | $2.89{ }^{-01}$ | $2.84{ }^{-01}$ | $2.80^{-01}$ | $2.77^{-01}$ | $2.74{ }^{-01}$ | $2.71^{-01}$ | $2.68{ }^{-01}$ |
| 1 | 78 | $7.20^{-01}$ | $6.82{ }^{-01}$ | $6.60^{-01}$ | $5.07{ }^{-01}$ | $4.46{ }^{-01}$ | $4.13{ }^{-01}$ | $3.91{ }^{-01}$ | $3.82{ }^{-01}$ | $3.72^{-01}$ |
|  |  | $3.67{ }^{-01}$ | $3.54{ }^{-01}$ | $3.49^{-01}$ | $3.46{ }^{-01}$ | $3.44{ }^{-01}$ | $3.42{ }^{-01}$ | $3.40^{-01}$ | $3.38{ }^{-01}$ | $3.36{ }^{-01}$ |
| 1 | 83 | $2.16{ }^{+00}$ | $2.18{ }^{+00}$ | $2.19^{+00}$ | $2.35{ }^{+00}$ | $2.49^{+00}$ | $2.63{ }^{+00}$ | $2.78{ }^{+00}$ | $2.88{ }^{+00}$ | $3.02^{+00}$ |
|  |  | $3.11^{+00}$ | $3.53{ }^{+00}$ | $3.89{ }^{+00}$ | $4.22^{+00}$ | $4.51^{+00}$ | $4.78{ }^{+00}$ | $5.03{ }^{+00}$ | $5.27{ }^{+00}$ | $5.49^{+00}$ |
| 1 | 84 | $3.27{ }^{+00}$ | $3.30^{+00}$ | $3.32{ }^{+00}$ | $3.58{ }^{+00}$ | $3.80{ }^{+00}$ | $4.01^{+00}$ | $4.24{ }^{+00}$ | $4.38{ }^{+00}$ | $4.59^{+00}$ |
|  |  | $4.72^{+00}$ | $5.32+00$ | $5.85{ }^{+00}$ | $6.32^{+00}$ | $6.75{ }^{+00}$ | $7.15{ }^{+00}$ | $7.51{ }^{+00}$ | $7.86{ }^{+00}$ | $8.18^{+00}$ |
| 1 | 85 | $4.37{ }^{+00}$ | $4.41^{+00}$ | $4.44^{+00}$ | $4.76{ }^{+00}$ | $5.05{ }^{+00}$ | $5.32^{+00}$ | $5.63+00$ | $5.83{ }^{+00}$ | $6.11^{+00}$ |
|  |  | $6.28{ }^{+00}$ | $7.07{ }^{+00}$ | $7.74{ }^{+00}$ | $8.33+00$ | $8.85{ }^{+00}$ | $9.33^{+00}$ | $9.77^{+00}$ | $1.02^{+01}$ | $1.05{ }^{+01}$ |
| 1 | 87 | $8.41^{-02}$ | $8.75{ }^{-02}$ | $8.93{ }^{-02}$ | $9.78{ }^{-02}$ | $9.96{ }^{-02}$ | $1.01^{-01}$ | $1.02^{-01}$ | $1.02^{-01}$ | $1.03^{-01}$ |
|  |  | $1.04{ }^{-01}$ | $1.07{ }^{-01}$ | $1.10^{-01}$ | $1.12{ }^{-01}$ | $1.14{ }^{-01}$ | $1.15{ }^{-01}$ | $1.16{ }^{-01}$ | $1.17^{-01}$ | $1.17{ }^{-01}$ |
| 1 | 88 | $1.09^{-01}$ | $1.10^{-01}$ | $1.10^{-01}$ | $1.19{ }^{-01}$ | $1.29{ }^{-01}$ | $1.38{ }^{-01}$ | $1.45{ }^{-01}$ | $1.49^{-01}$ | $1.54{ }^{-01}$ |
|  |  | $1.56{ }^{-01}$ | $1.64{ }^{-01}$ | $1.69^{-01}$ | $1.73{ }^{-01}$ | $1.75{ }^{-01}$ | $1.77^{-01}$ | $1.78{ }^{-01}$ | $1.78{ }^{-01}$ | $1.79^{-01}$ |
| 1 | 90 | $7.10^{-02}$ | $7.17^{-02}$ | $7.22^{-02}$ | $7.47{ }^{-02}$ | $7.64{ }^{-02}$ | $7.87{ }^{-02}$ | $8.19{ }^{-02}$ | $8.42^{-02}$ | $8.76{ }^{-02}$ |
|  |  | $8.98{ }^{-02}$ | $9.96{ }^{-02}$ | $1.07{ }^{-01}$ | $1.13{ }^{-01}$ | $1.17^{-01}$ | $1.20^{-01}$ | $1.23{ }^{-01}$ | $1.25{ }^{-01}$ | $1.26{ }^{-01}$ |
| 1 | 91 | $1.13{ }^{-01}$ | $1.13{ }^{-01}$ | $1.13{ }^{-01}$ | $1.18{ }^{-01}$ | $1.25{ }^{-01}$ | $1.32{ }^{-01}$ | $1.39^{-01}$ | $1.44{ }^{-01}$ | $1.49^{-01}$ |
|  |  | $1.52^{-01}$ | $1.64{ }^{-01}$ | $1.72^{-01}$ | $1.77{ }^{-01}$ | $1.81^{-01}$ | $1.83{ }^{-01}$ | $1.84{ }^{-01}$ | $1.85{ }^{-01}$ | $1.85{ }^{-01}$ |

(This table is available in its entirety in a machine-readable form in the online journal. A portion is shown here for guidance regarding its form and content.)
$3 d^{4} 4 s{ }^{6} D_{7 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{6} D_{5 / 2}^{o}$ transition in the upper plot. Clearly further investigation into the oscillator strengths for these transitions is necessary.

Finally, in Figures 7-9 we present the collision strength as a function of incident electron energy, the corresponding effective collision strength as a function of electron temperature, and the reduced energy plots of Burgess \& Tully (1992), respectively, for the $3 d^{4} 4 s{ }^{4} D_{3 / 2}^{e} \rightarrow 3 d^{4} 4 p{ }^{4} P_{1 / 2,3 / 2,5 / 2}^{o}(8-86,8-87$, and 8-92) transitions. Agreement with the collision strengths of Bautista et al. (2009) in Figure 7 is disappointing, with their predictions lying substantially higher than the present collision strengths for two out of the three transitions. The Maxwellian averaged effective collision strengths derived from these collision strengths and displayed in Figure 8 lie about a factor of two higher at all temperatures considered. The agreement for the $3 d^{4} 4 s{ }^{4} D_{3 / 2}^{e} \rightarrow 3 d^{4} 4 p^{4} P_{5 / 2}^{o}$ transition is, however, excellent across the temperature range. This would suggest that there is considerable mixing between these finestructure levels. The source of the differences may also stem from the differing representations for the $3 d^{4} 4 s$ states in the two calculations. Finally, in Figure 9 we present the reduced energy plots for these transitions and predict the high-energy limits from a variety of $f$-value sources. We see from Table 2 that the present data are in excellent agreement with the most recent experimental work of Nilsson et al. (2006) for all three transitions of interest. From Figure 9 the present data are as expected in closest agreement with the predictions of Nilsson et al. (2006).

It is clearly evident from Table 2 that there is a paucity of data currently available in the literature for the oscillator strengths for transitions among the fine-structure levels of Cr II. These data will be of significant use to astrophysicists and plasma physicists in their diagnostic applications and for this reason we present in Table 3 a list of oscillator strengths $(f)$ and transition
probabilities (A) for all dipole allowed (E1) transitions among the lowest 274 states (removing the final six unobserved states) of $\mathrm{Cr}_{\text {II }}$ consisting of configurations $3 d^{5}, 3 d^{4} 4 s$, and $3 d^{4} 4 p$. The relativistic configuration-interaction CIV3 code of Hibbert (1975) was utilized in the evaluation of this data set and the $J \pi$ selection rules were adopted. It should be noted that the agreement between length and velocity oscillator strengths is really quite good for the majority of the transitions considered, including those which are significantly weak and the oscillator strength is very small. This is a good indicator of the suitability and accuracy of the model used in the present calculation.
Finally, in Table 4 we tabulate the effective collision strengths computed in the present $280 j j$ approximation for all of the Cr II allowed transitions considered in the present study. This table will compliment the forbidden line tabulation of Wasson et al. (2010). The transitions in Tables 2-4 are labeled according to the index values assigned in Table 1. Temperatures ranging from $T=2000-100,000 \mathrm{~K}$ are included in the tabulation.

## 4. CONCLUSIONS

In this present work, we have computed collision strengths and Maxwellian averaged effective collision strengths for 8669 allowed transitions among the fine-structure levels of Cr II. This approximation contains all levels associated with even configurations $3 d^{5}$ and $3 d^{4} 4 s$ and the odd configuration $3 d^{4} 4 p$, leading to a substantial 280jj level, 1932 coupled channel scattering problem. The astrophysically significant effective collision strengths have been calculated for an extensive temperature range of 2000-100,000 K, extending the range of data previously available.

Comparison with the work of Bautista et al. (2009) has led to a few discrepancies. Looking at the first optically allowed transitions from the ground state shows promising agreement between the two calculations. Further investigation, however,
highlights a number of instances where the data of Bautista et al. (2009) predict collision strengths higher than the present work. This inevitably has a knock on effect on the effective collision strengths produced. Despite these differences, the strong agreement observed in many instances shows the close correlation between these two works.

The results presented here signify a large increase in the size and sophistication of the model used and represent the largest scattering calculation performed for Cr II. Care and precision has been taken in building the target model, in ensuring convergence of the collision strengths, particularly for higher incident electron energies and as an added check the reduced collision strengths have shown that the infinite-energy limits are in very good agreement with experimental predictions. Comparison with other theoretical models and experimental calculations have in many instances corroborated our results.

Conclusive assessment of the accuracy of the effective collision strengths presented here still remains difficult, especially since we only have the work of Bautista et al. (2009) with which to compare. The definitive testing of these data will come from subsequent application of these results within the disciplines of astrophysics or plasma physics.

All of the data presented in this paper are available from the authors on request including collision strengths, effective collision strengths, and oscillator strengths. Alternatively, the reader is directed to our Web site http://web.am.qub.ac.uk/apa where all the data are freely available for download.

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